R C T U Q P U EMSI

Full-Scale Treatment of 1,4-Dioxane Using a Bioreactor

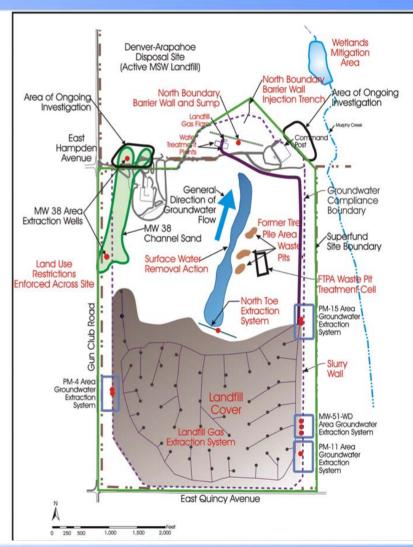
Tim Shangraw, P.E., EMSI William Plaehn, P.E., Parsons December 6, 2006

BACKGROUND

- Lowry Landfill Superfund Site
 - Located just outside Denver, Colorado
 - A 480 acre industrial and municipal landfill from the mid-1960's to 1980
 - Estimated 138 million gallons of liquid industrial wastes disposed into unlined pits until 1980
 - Placed on the Superfund list in 1984
 - Record of Decision lodged in 1994
 - Containment of groundwater is primary objective

SITE MAP

- Water Treatment Plant (WTP) to treat recovered groundwater
- Two primary waste streams
 - North Boundary Barrier Wall (NBBW)
 - 7 to 10 gpm
 - Relatively few organics
 - North Toe Extraction System (NTES)
 - 0.5-3 gpm
 - Flowrate determined by trench level
 - High concentration of organics



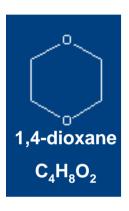
OPERATIONS

- WTP Pre-treats extracted groundwater prior to discharging to the Denver Metro Water Reclamation Facility
- Treatment includes:
 - Equalization
 - Chemical Softening
 - pH Adjustment
 - Bag Filtration
 - UV/Oxidation
 - Activated Carbon

PERFORMANCE

- Treatment train cannot remove 1,4-dioxane to below permit (2.0 ppm) due to poor UV transmittance, hydroxyl scavenging with NTES waters, and poor sorption on GAC
- Other permit parameters can be met by GAC alone
- Evaluated a number of technologies:
 - Chemical precipitation (to remove UV-Oxidation interferences)
 - Additional Advance Oxidation Processes
 - Thermal Treatment
 - Biological treatment

LITERATURE SEARCH





- Reactors operated at 35°C
- Attached growth bio-reactor
- 1,4-dioxane removal as sole substrate
- Reaction efficiency decreased sharply with temperature

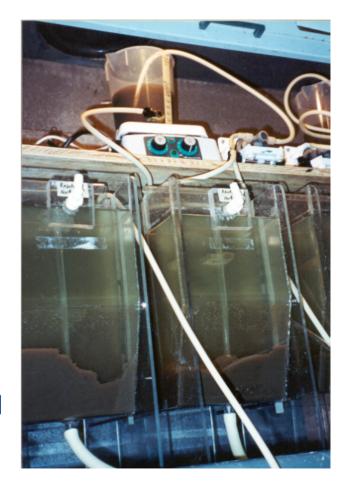
NC State

- Reactors operated at 35°C
- Used RBC for seed reactor, performed batch tests with seed
- 1,4-dioxane removal as cometabolite with Tetrahydrofuran (THF)
- 1,4-dioxane and THF are structural analogues
- THF acts as a competitive inhibitor of 1,4-dioxane degradation
- Reaction rates decreased sharply with temperature



BENCH STUDY

- Two feeds tested:
 - NTES
 - 30% NTES, 70% NBBW Blend
- Operated at room temperature (18-25°C)
- Blended water showed more favorable results
 - NTES 53-77% 1,4-dioxane/THF reductions
 - Blended 73-88% 1,4-dioxane/THF reductions
- Nitrification was not complete using NTES waters only allowing nitrite accumulation and probable ammonia toxicity



PILOT STUDY

- Parallel 300 Gallon, fixed film, Moving Bed Bio-Reactors (MBBRs)
- Acclimation started September 2002
- Two reactors to study effect of temperature, Reactor 1 @ 25°C and Reactor 2 @ 15°C



STUDY CONCLUSIONS

- Biological degradation of 1,4-dioxane is possible with blended stream
- 1,4-dioxane and THF degradation was promoted by the indigenous bacteria population
- 1,4-dioxane was degraded to greater than 95% efficiency consistently at organic loadings (F/M ratio) of 0.04 to 0.075 g D&T COD/g TS*d
 - F = food as grams of COD equivalent of 1,4-dioxane and THF
 - M = mass of solids on media in reactor
- Reductions occurred in 12 to 15 hours at F/M ratios noted
- Biodegradation of 1,4-dioxane at 15°C was possible but more susceptible to upset conditions
- 1,4-dioxane was not removed by volatilization
- THF was required for 1,4-dioxane degradation
- Instantaneous increases in loadings of 25-percent resulted in 2-3 days decreased 1,4-dioxane degradation efficiency
- Move forward with full-scale biological treatment of 1,4-dixoane

PROCESS WATER CHARACTERISTICS

Parameter	30:70 NTES:NBBW Blend
Flow	1.3–6.0 gpm
1,4-dioxane	13,000-25,000 μg/L
Tetrahydrofuran (THF)	30,000-60,000 μg/L
Total Suspended Solids (TSS)	75-170 mg/L
Ammonia	140-230 mg-N/L
Nitrite	<0.1-1.7 mg-N/L
Nitrate+Nitrite	14-21 mg-N/L

DESIGN AND OPERATION

- Design criteria:
 - Flowrates up to 10 gpm blended flow
 - Effluent water quality < 1,000 μg/L 1,4-dioxane
- Slip-stream from main WTP
- 300 gallon Sedimentation Tank for coarse solids sedimentation
- 300 gallon Blend Tank for blending, pH adjustment, and phosphorous addition
- 3 aerobic, fixed-film, MBBRs
 - 2 x 3,900 gallons; 1 x 5,400 gallons
 - Kaldnes[®] media used for fixed film growth
 - 32 to 36% of reactor volume filled with media or approximately 8,400 m² of surface area
- Coarse bubble diffusers for aeration and mixing

DESIGN AND OPERATION (cont)

DECICITATION CI EIGHT (COIII)

3 MBBRs **Sedimentation Tank** O)B-3331 - Acid/Coustic Addition T-2220 P-2231 Phosphorous Addition Acid/Coastic Addition-P-2211 M Slip Stream Feed (see Figure 3.1a) T-3331 Acid/Coastic Addition P-2232 Recycle Sedimentation Tanb Blend Tank T-3310 T-3320 Influent Blending Solidsto Sludge olidsto Sludge Holding Tanks (T-351/352) Holding Tanks (T-351/352) Overflow To T-3340 Discharge (See Figure 3 la) Acid/Caustic Addition **Blend Tank Effluent Tank**





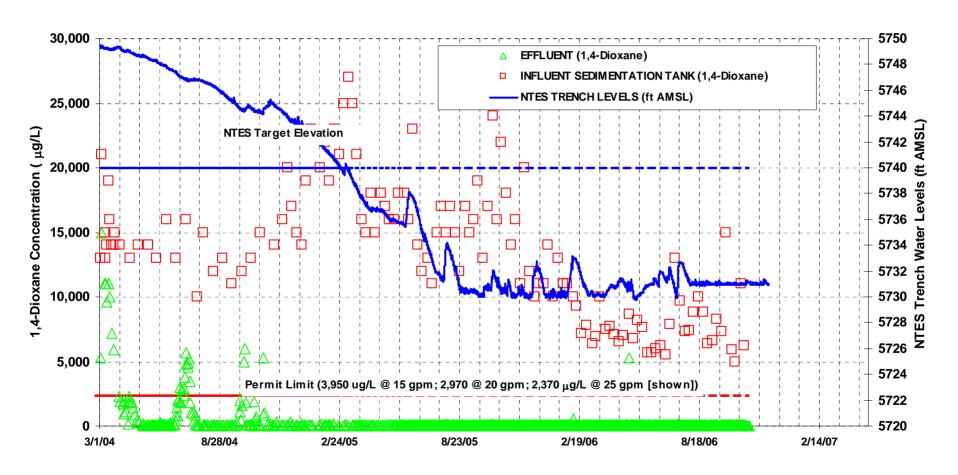




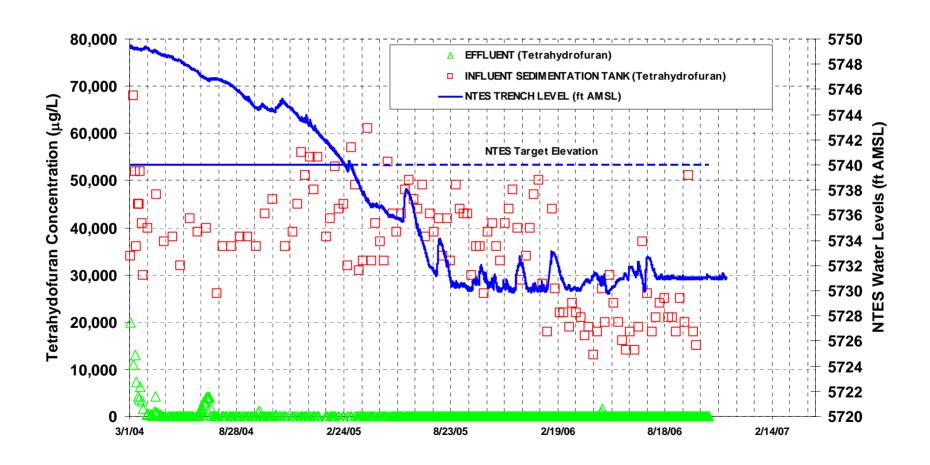
RESULTS

- Acclimation began in March 2004
 - Seeded with indigenous bacteria inherent to the raw NTES stream
 - Low flow-rate (1.3 gpm)
- Transition began in May 2004
 - Flowrates between 2.7 and 6.0 gpm
- Full-Scale operations began in January 2005
 - Maximum flowrate of 6.0 gpm dictated by flowrate from NTES not based on system capability
 - Successful compliance testing

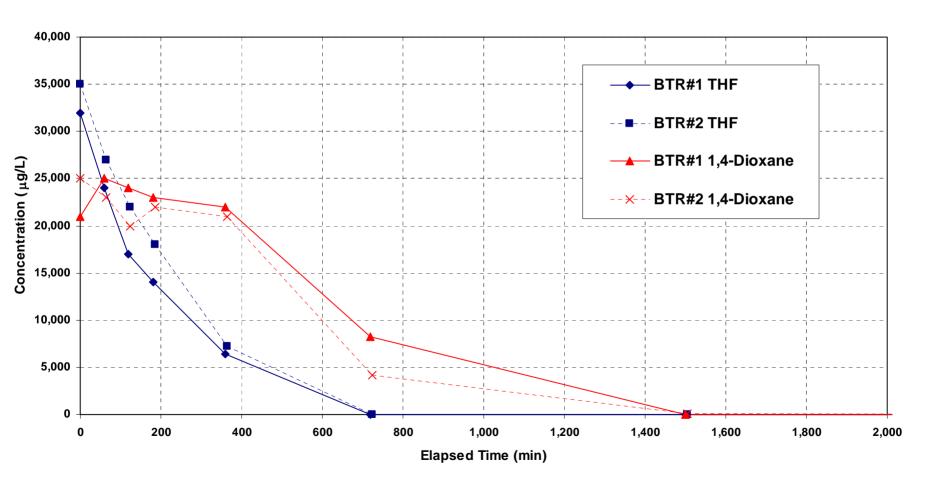
1,4-DIOXANE



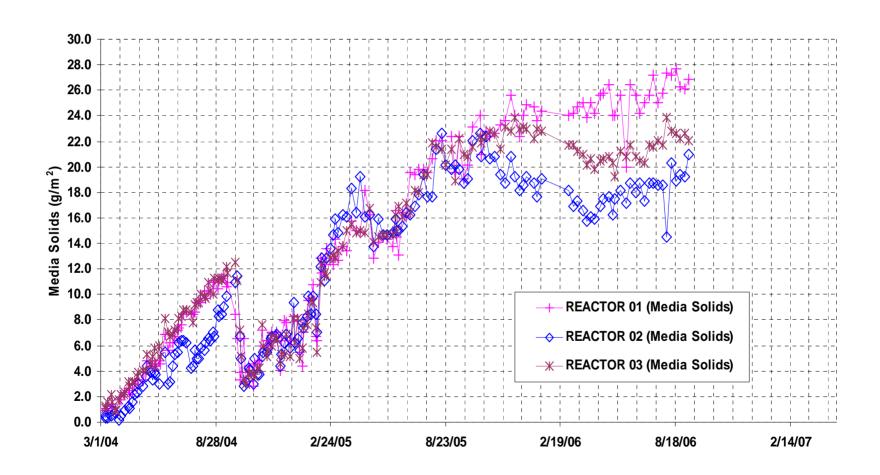
TETRAHYDROFURAN



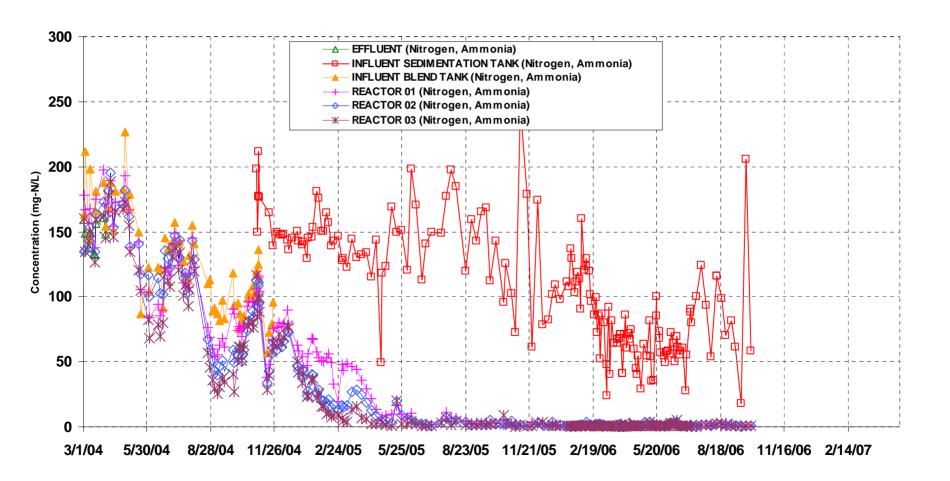
EXAMPLE DEGRADATION PROFILE



MEDIA SOLIDS



AMMONIA



CONCLUSIONS

- Indigenous bacteria successful at degradation of 1,4-dioxane and THF
- System capable of treating 6 gpm of 25,000 μg/L 1,4-dioxane to less than 1 μg/L (>99.9% removal efficiency)
- System capable of treating 6 gpm of 60,000 μg/L THF to less than 1 μg/L (>99.9% removal efficiency)
- Organic loading (F/M ratio) between 0.04 and 0.07 g D&T COD/g TS*d resulted in high removal efficiencies
- Nitrification is occurring with no inhibitory effect on 1,4-dioxane degradation
- Temperature control (at approximately 23°C) is necessary to maintain high removal efficiencies
- No longer need UV Oxidation system because of robustness of biological system
- Other VOCs in process waters are treated in the biological system

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- Dr. Craig Adams U. of Missouri

QUESTIONS?

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