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Overview of Remediation Technologies for Radionuclides in Soil and Groundwater

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Remediation technologies operate at the intersection of

- radionuclide characteristics
- the target problem
- remedy functionality
- remediation objectives





- Radionuclide characteristics related to remediation
- Considering end states and attenuation in remedy decisions
- Remedy technologies and approaches
- Remedy implementation
- Discussion focused on
 - Uranium, Tc-99, Sr-90, I-129, tritium
 - Groundwater protection and groundwater remediation

Radionuclide Characteristics (Friend or Foe)



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Half-life

Shorter is better (when exposure is controlled)

Sr-90 or tritium compared to uranium, I-129, or Tc-99

- Mobility (sorption)
 - Very low mobility generally good
 - Medium or high mobility depends on the situation
 - Attenuated transport can be helpful (vadose zone contamination) or problematic (P&T)
 - Secondary sources are problematic unless balanced by attenuation

Radionuclide Characteristics (Friend or Foe)



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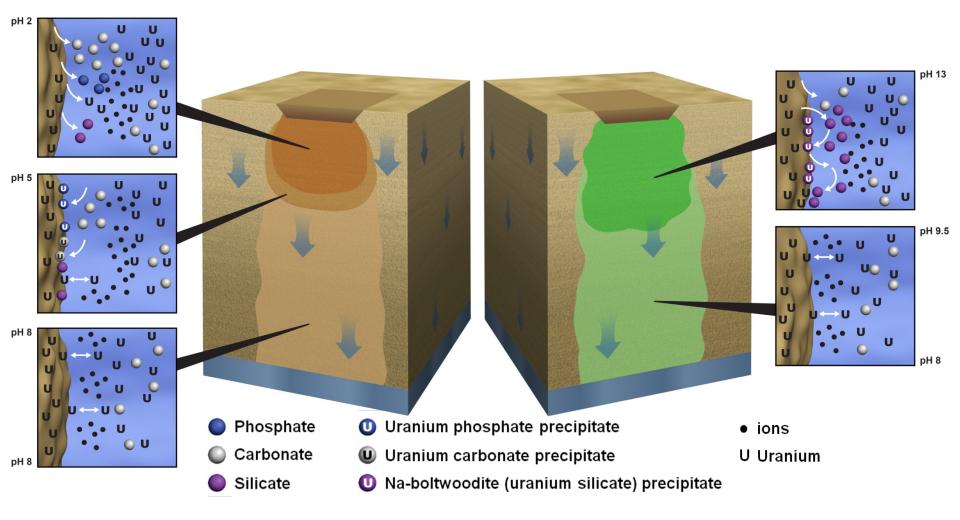
Biogeochemical interactions

- Helpful
 - Uranium and Sr-90 interactions with phosphate
 - Uranium silicate precipitates
- Mixed
 - Uranium and I-129 (and Cr) interactions with carbonate
 - Depends on location/extent
 - I-129 species transformation
 - Depends on change in mobility and potential for attenuation/sequestration
 - Uranium and Tc-99 redox
 - Depends on setting and role in a remedy
- No interactions
 - tritium

Disposal Chemistry



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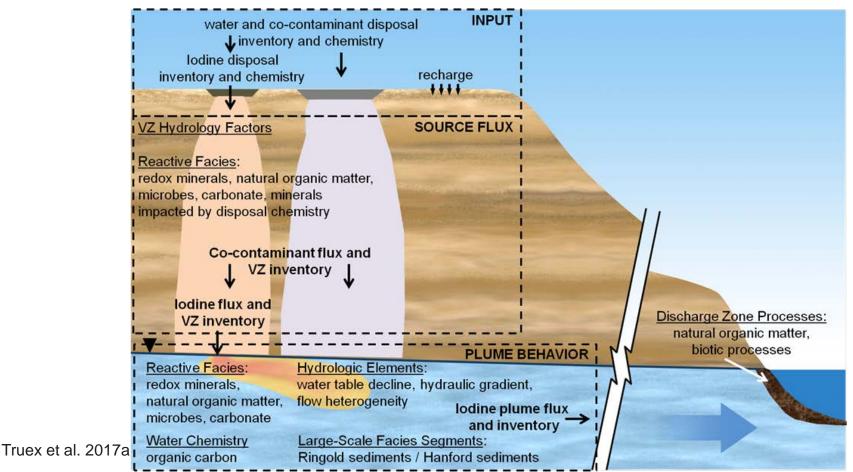
Szecsody et al. 2013 Truex et al. 2014

Radionuclide Characteristics (Friend or Foe)



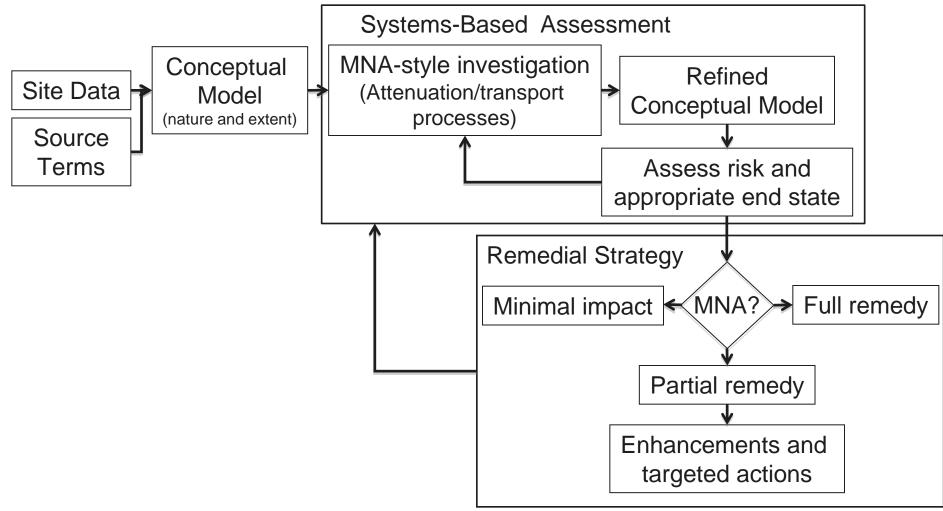
The Conceptual Site Model helps us decide:

- Friend or foe for risk and transport
- Friend or foe for remediation



Considering End States and Attenuation in Remedy Selection





Remedy Technologies and Approaches



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Vadose zone

Attenuation

Consider transport processes in the vadose zone

Flux control (enhanced attenuation)

Physical stabilization

Hydraulic control

Biogeochemical stabilization

Extraction (e.g., excavation, soil flushing)

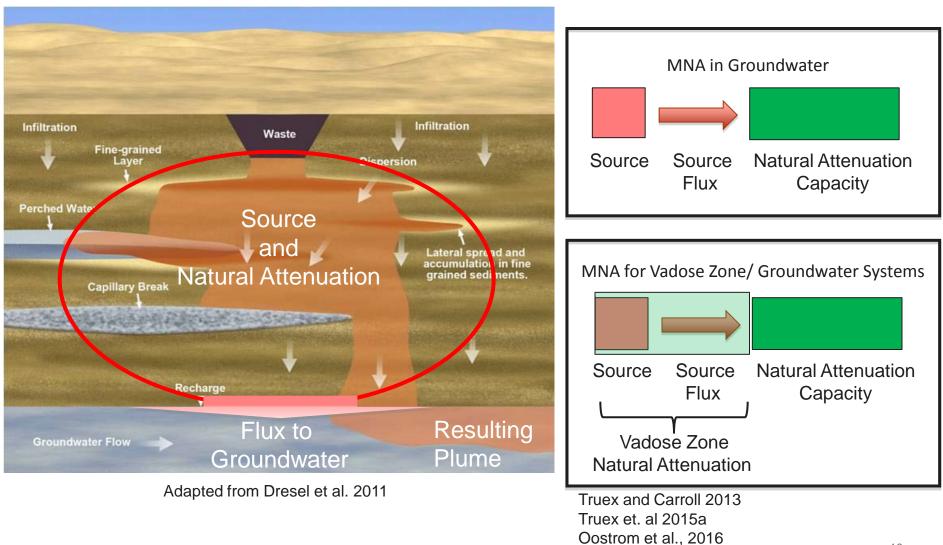
Cost/benefit

Groundwater treatment (e.g., phosphate)

 Consider vadose zone source characteristics for groundwater impact

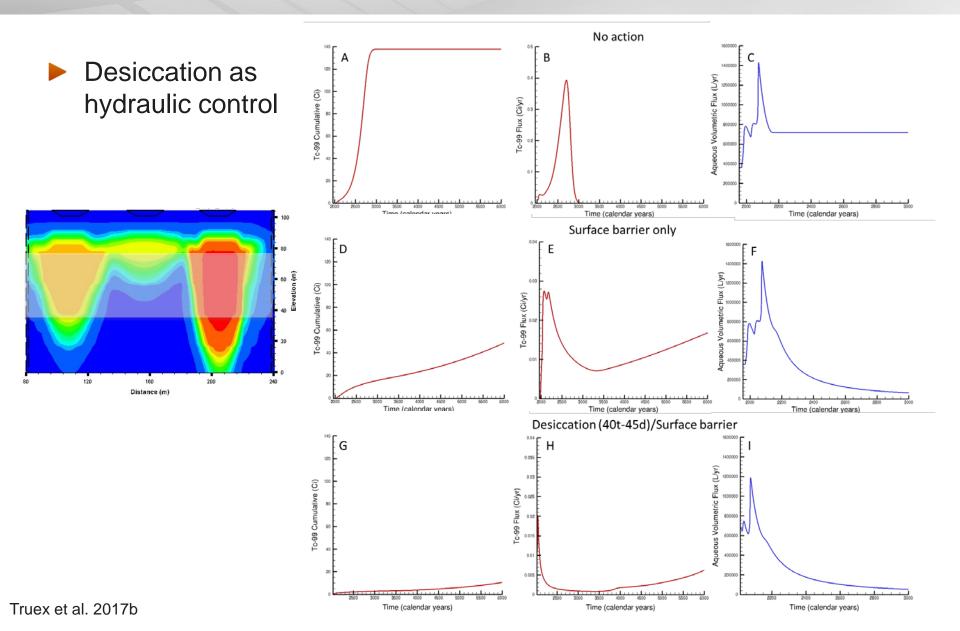
Attenuation





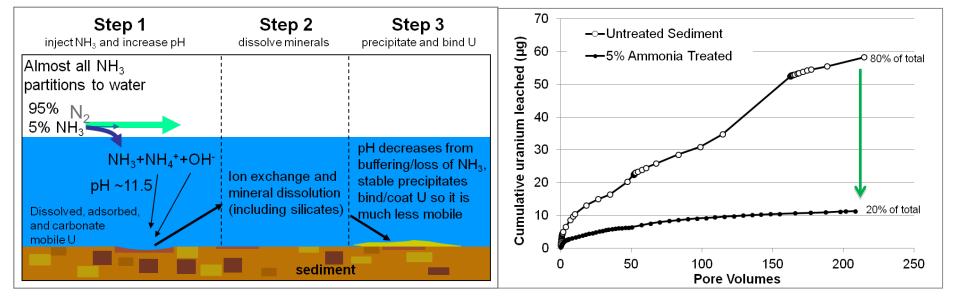
Desiccation





Geochemical stabilization – vadose zone

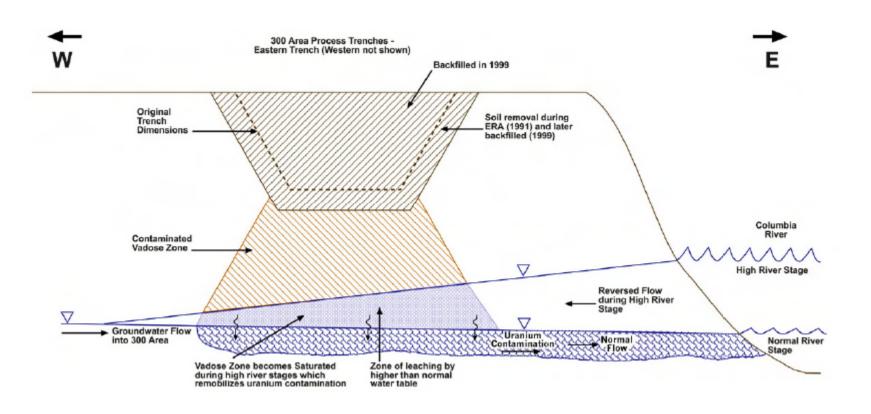
Ammonia gas for uranium sequestration





Uranium source zone

Periodically rewetted zone



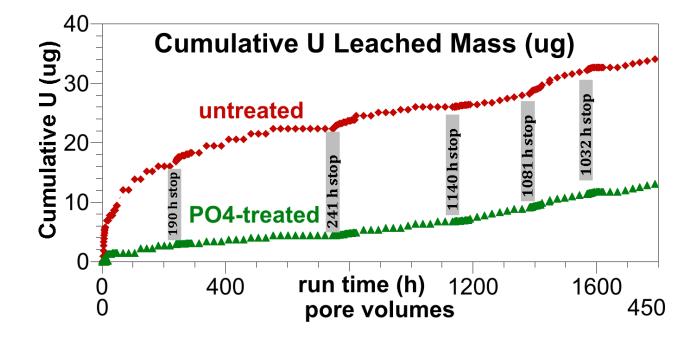
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Geochemical stabilization – periodically rewetted zone



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Phosphate treatment for uranium



Remedy Technologies and Approaches



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Groundwater

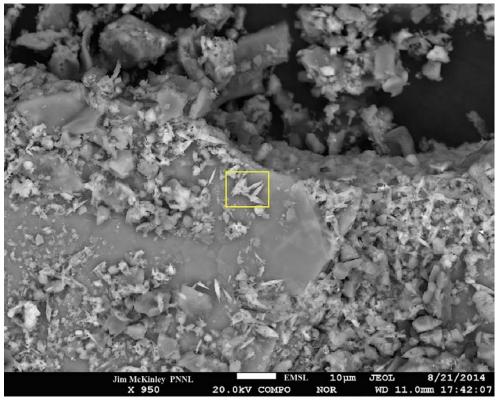
Attenuation

- EPA guidance
- Enhanced Attenuation and Source Control
 - Physical stabilization
 - Hydraulic control
 - Biogeochemical stabilization
- Extraction (P&T)
 - Cost/benefit
- Volumetric Treatment/Permeable Reactive Barriers
 - Scale, transport, attenuation

Carbonate interactions



Uranium, iodate, and chromate co-precipitates with calcite



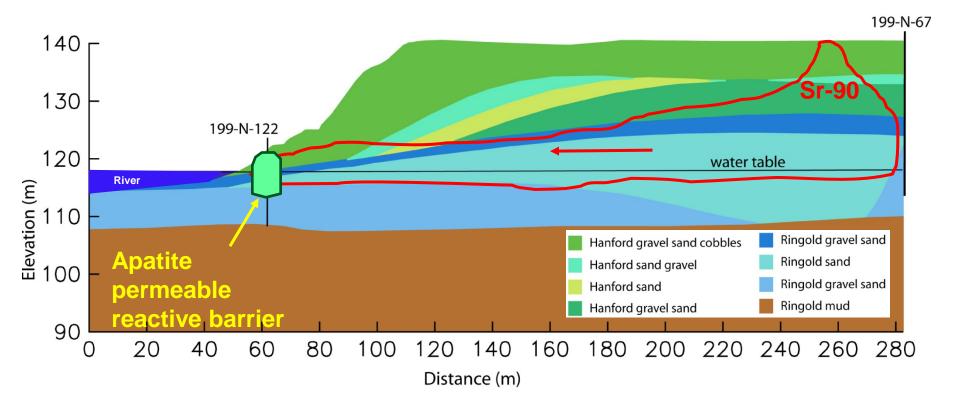
Cr-calcite observed in a Hanford field sediment

Truex et al. 2015b

100-N Strontium



- Only near-river strontium is a risk to the river
- Monitoring linked to remedy approach



Remedy Implementation



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Amendment distribution

- Vadose zone gas phase
- Phosphate mobility
- Particles
- Bioremediation amendments

Reductants



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ZVISMI



Truex et al. 2011a Truex et al. 2011b

Remedy Implementation



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Adaptive Site Management

- National Research Council
- ITRC
 - Remediation Management of Complex Sites
 - <u>http://rmcs-1.itrcweb.org/</u>
- Exit Strategies (P&T)
 - <u>http://bioprocess.pnnl.gov/Pump-and-Treat.htm</u>

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