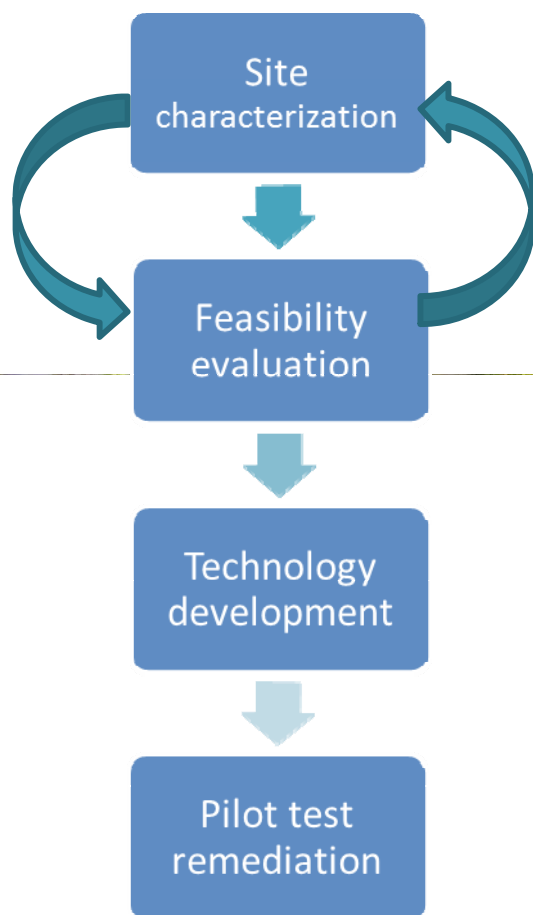


# Technologies for Biogeochemical and Hydrogeologic Characterization and Their Integration for Site Remediation



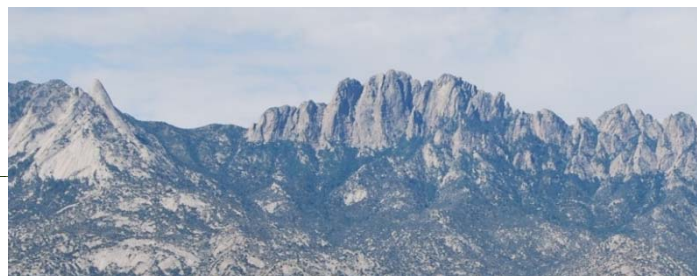
**Michelle M. Lorah**

U.S. Geological Survey

MD-DE-DC Water Science Center

[mmlorah@usgs.gov](mailto:mmlorah@usgs.gov)

- Complex hydrogeology
  - fractured rocks
  - low permeability layers; rock matrix



- Difficult contaminant mix
  - DNAPL
  - dissolved and highly sorbed mix
  - emerging contaminants

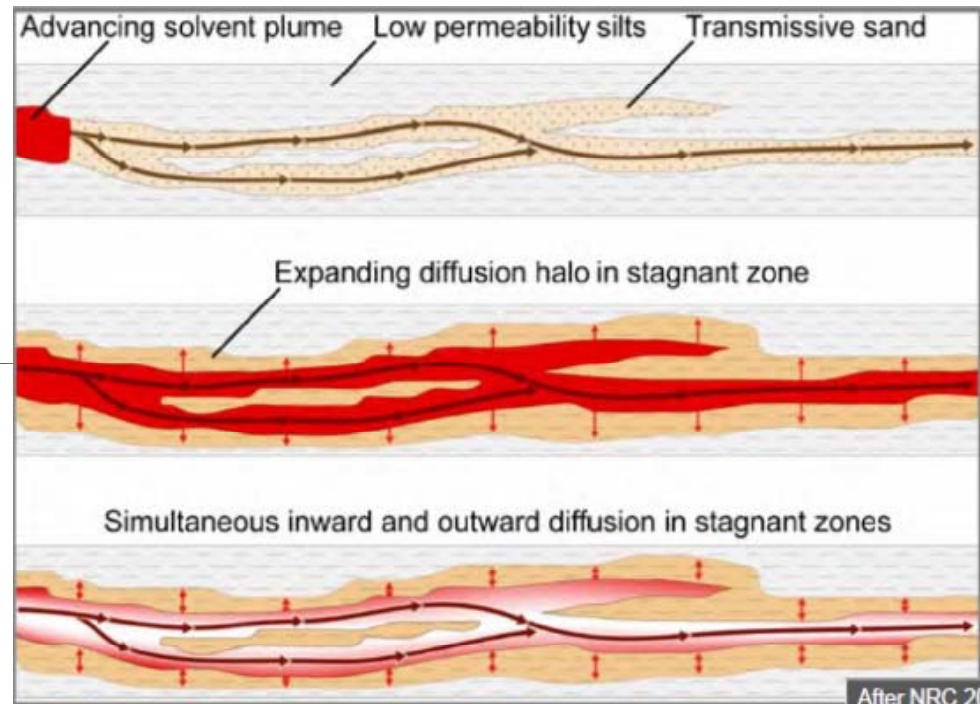


- Sensitive habitat or location
  - wetlands
  - bottom sediments

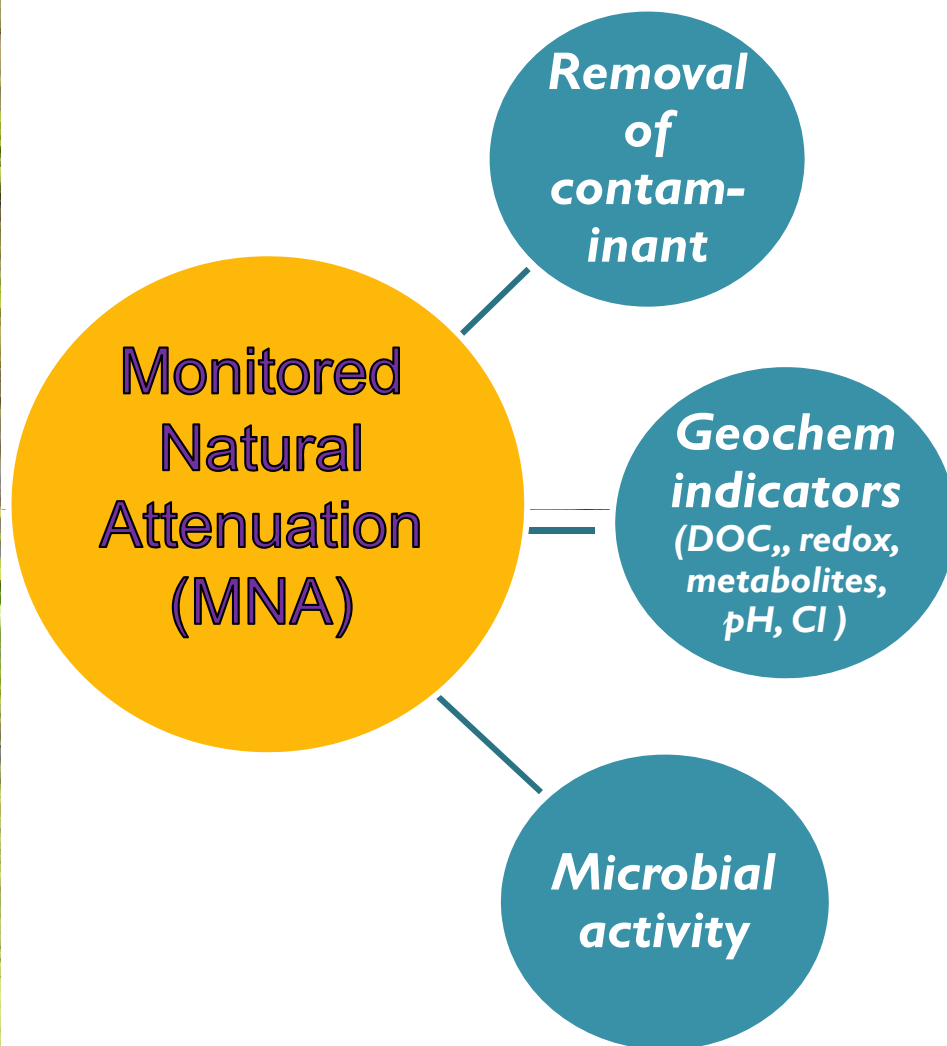


# Biogeochemical Characterization- Why?

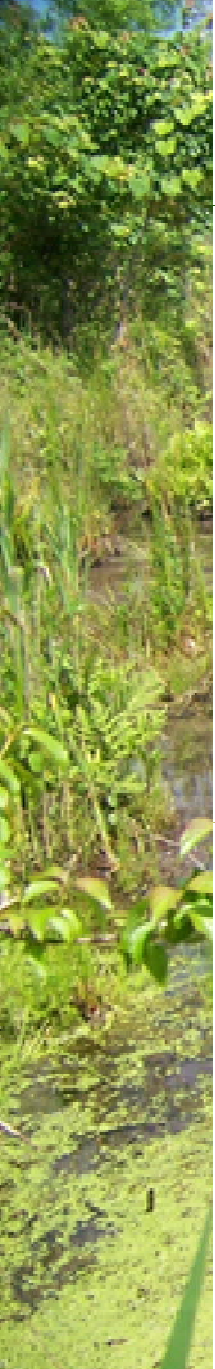
- Provide the remedy-  
MNA, bioremediation,  
biogeochemical reduction
- Secondary effects-  
alteration of natural  
biogeochemical conditions,  
or from presence of  
secondary contaminants
- Long-term efficiency
  - changes in transmissive  
plume with remediation
  - low permeability zones
  - “slow” processes key  
(back diffusion, sorption  
and desorption, abiotic  
and biotic degradation  
reactions)



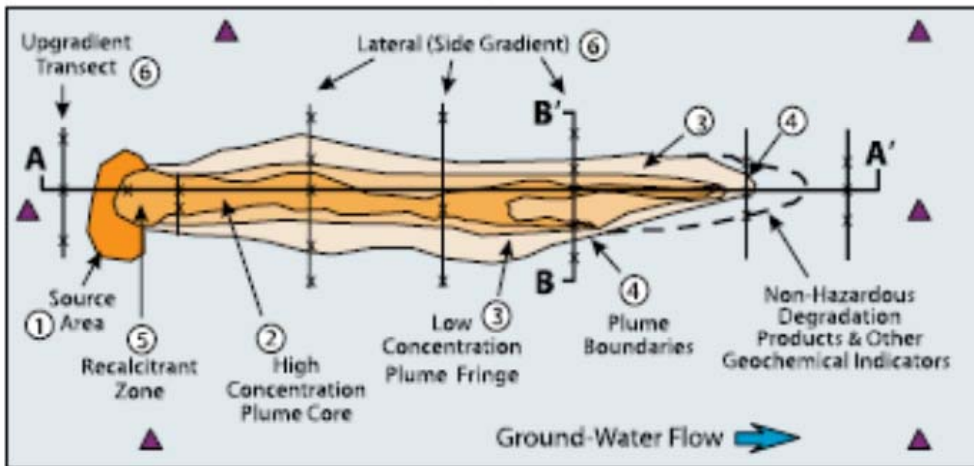
# Biogeochemical Characterization- How?



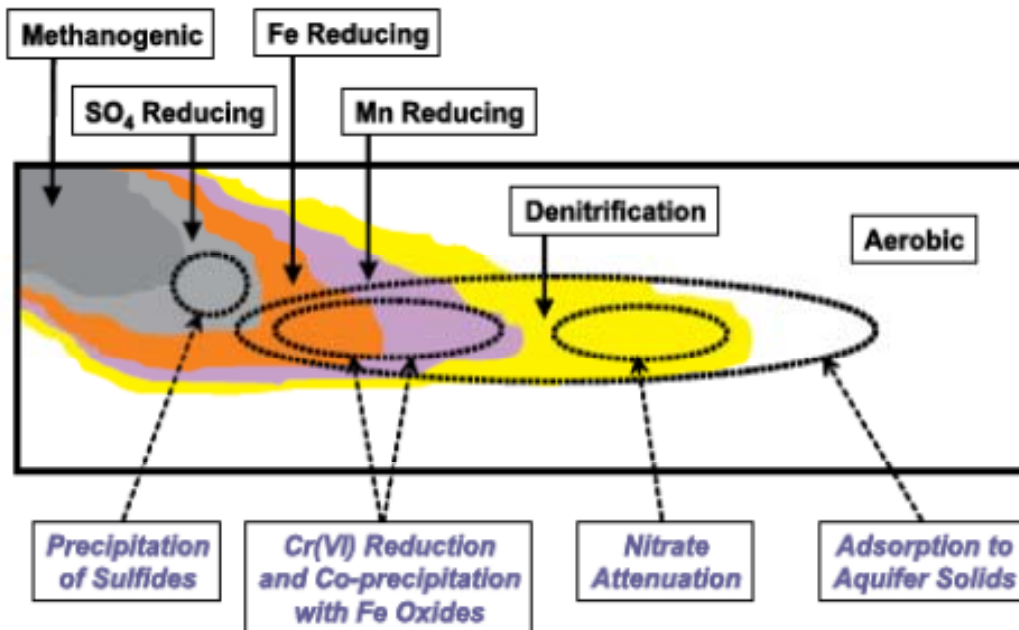
- MNA protocols provide good framework and tools
- Relevant protocols for organics, radionuclides, and non-radionuclide inorganic
- Three lines of evidence for chlorinated solvents



### Site (Map View)



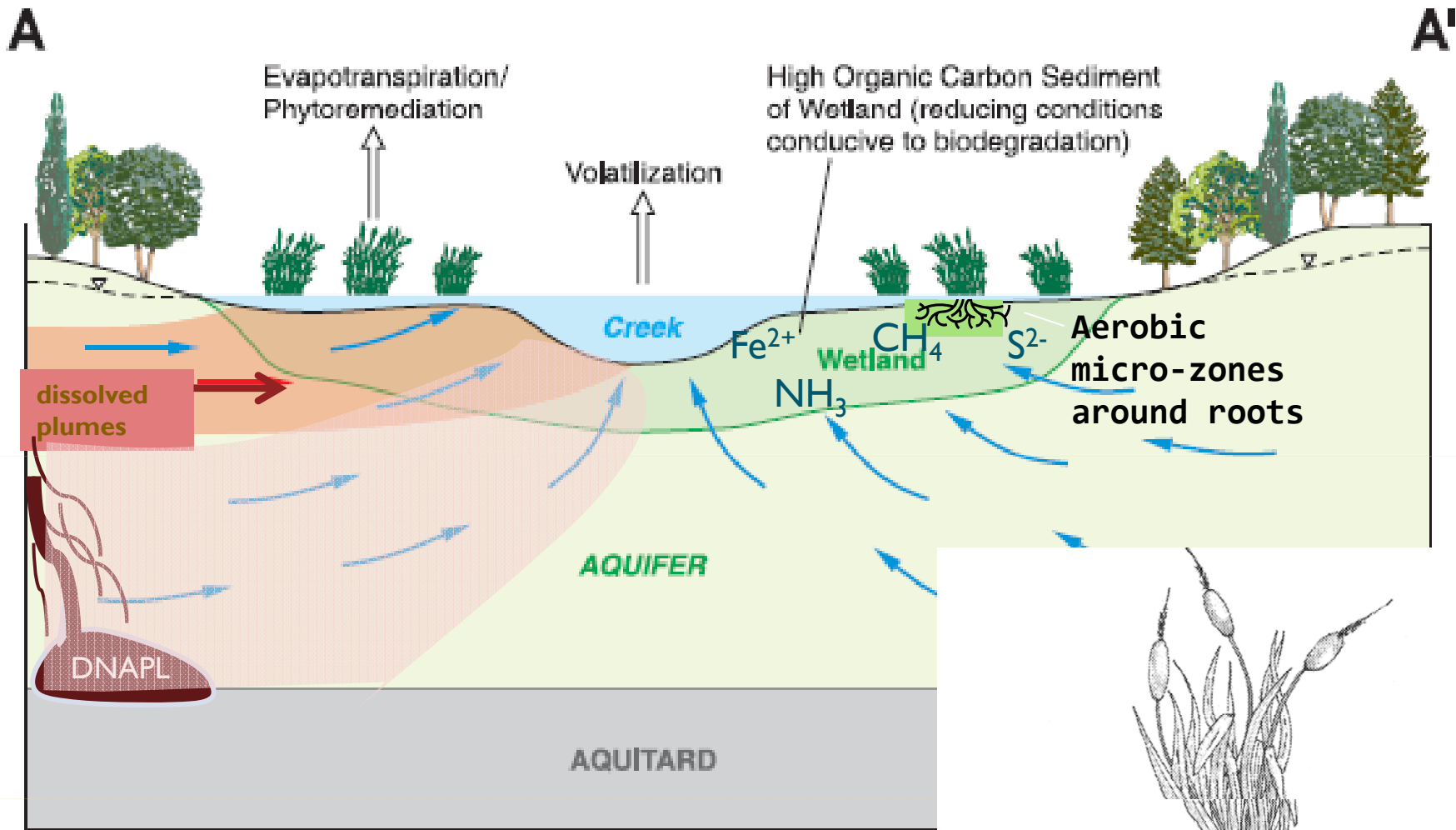
### Target Monitoring Zones



from *Monitored Natural Attenuation of Inorganic Contaminants in Ground Water, Volumes 1 and 2*

## Considerations

- History and stage of plume evolution
- DNAPL or LNAPL presence
- Sample key parts of plume, including “transition zones”
- Multilevel sampling-high resolution
- Spatial and temporal variability
- Interaction with and formation of solids



## WETLANDS- LARGE TRANSITION ZONE (MODIFIED FROM LORAH ET AL., 2005)

Fig. 1. Possible interactions in the root zone of wetlands for wastewater treatment.

# Canal Creek Area, Aberdeen Proving Ground

## Chlorinated VOCs- Anaerobic degradation

### Parent Contaminants

Chlorinated ethanes:

HCA= hexachloroethane

PtCA= pentachloroethane

1122TeCA= 1,1,2,2-tetrachloroethane

Chlorinated ethenes:

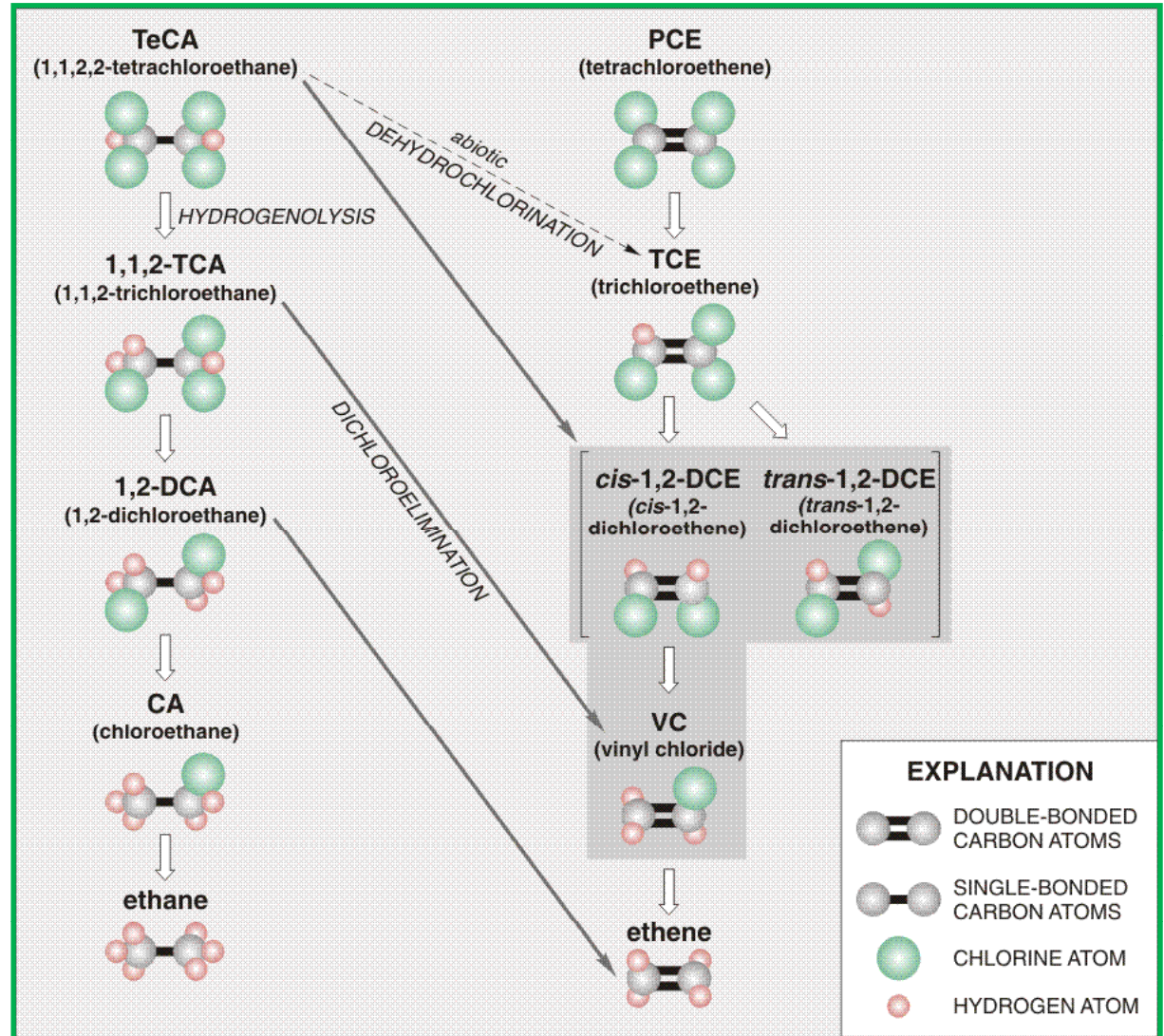
PCE= tetrachloroethene

TCE= trichloroethene

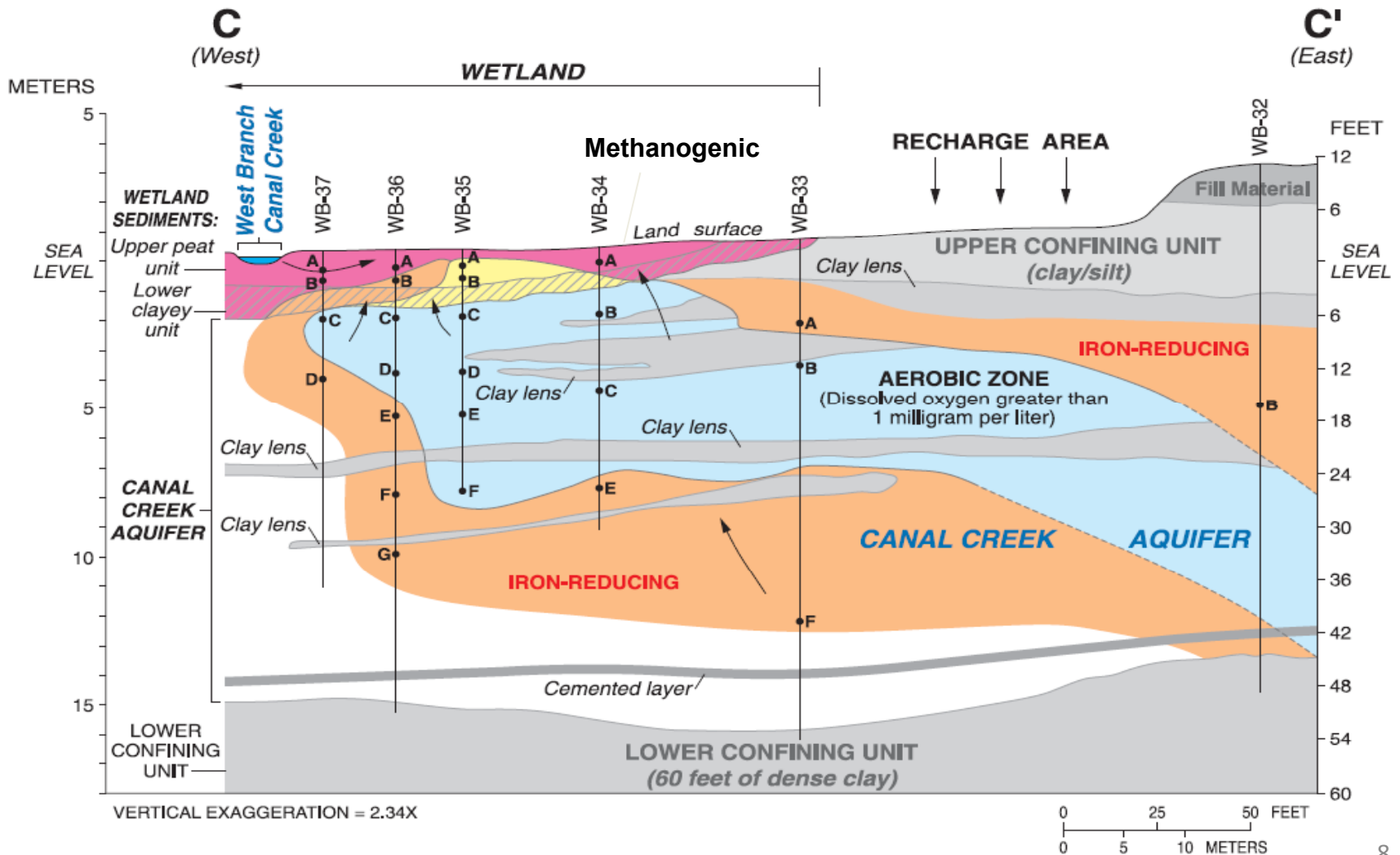
Chlorinated methanes:

CT= carbon tetrachloride

CF= chloroform

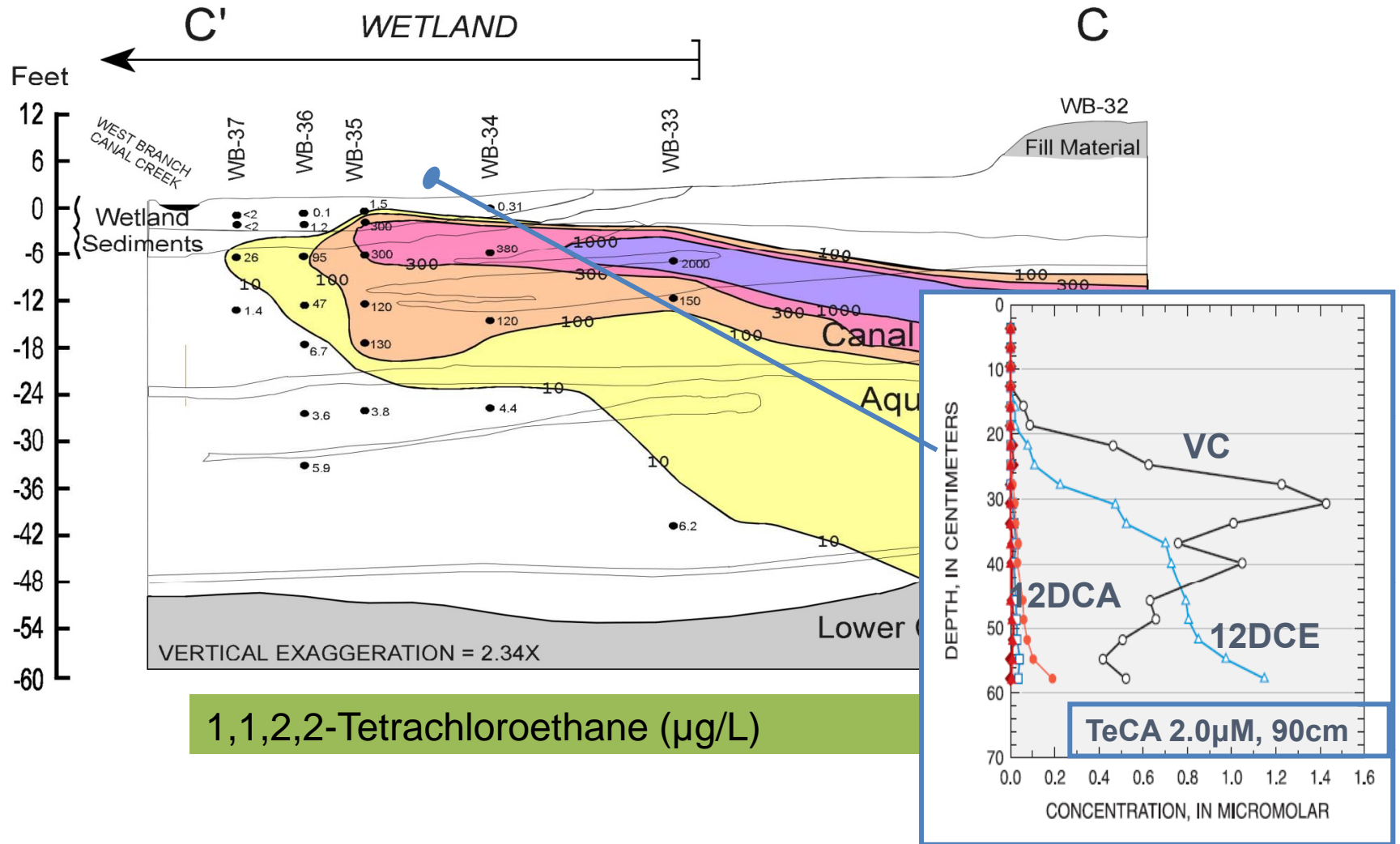


# West Branch Canal Creek, Natural Attenuation Study Area: Redox

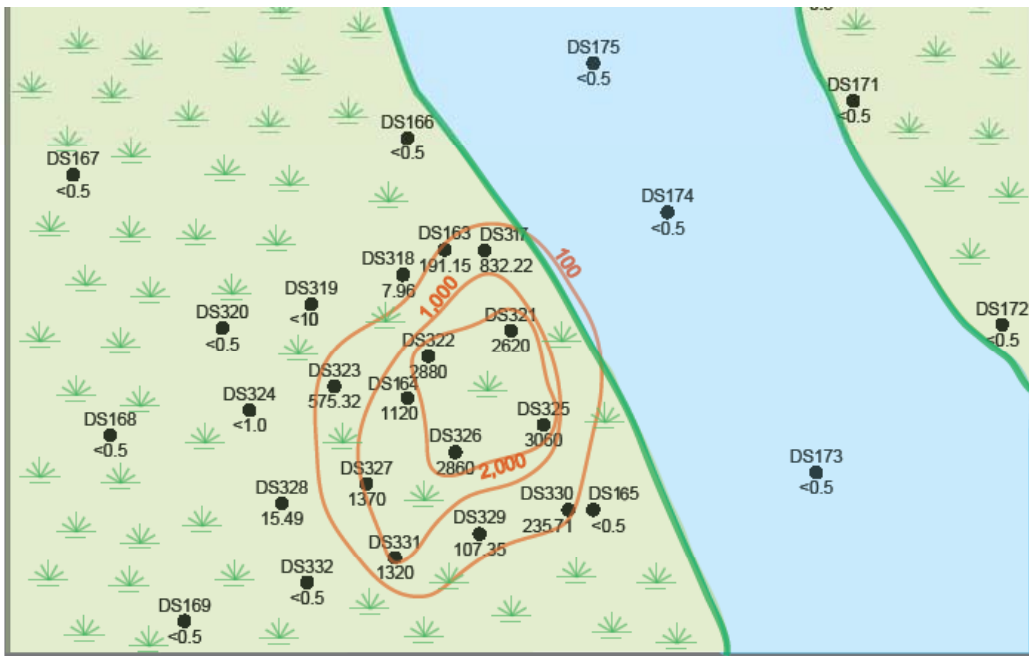
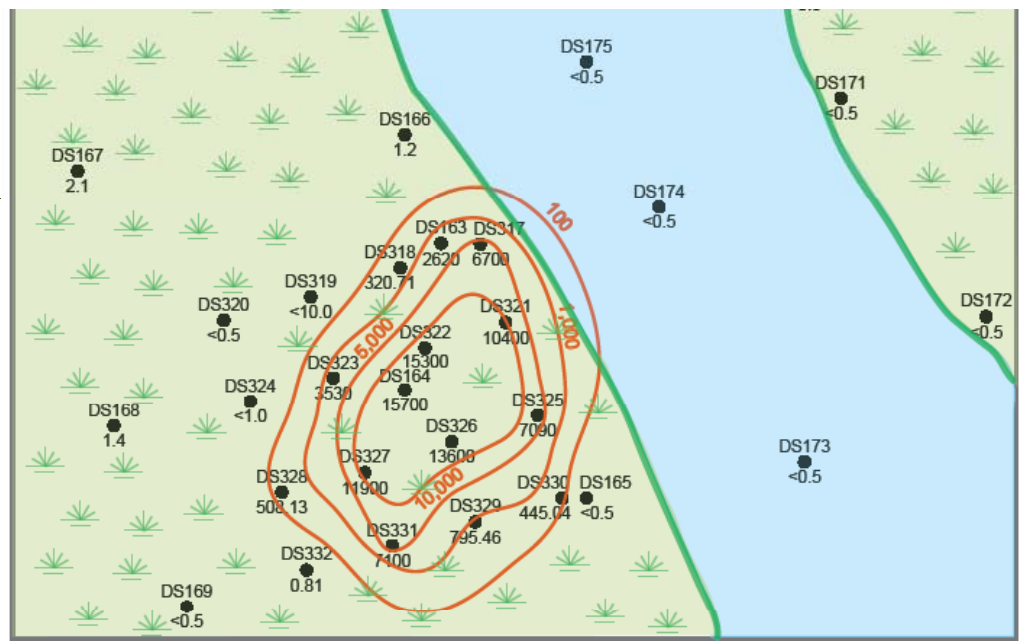
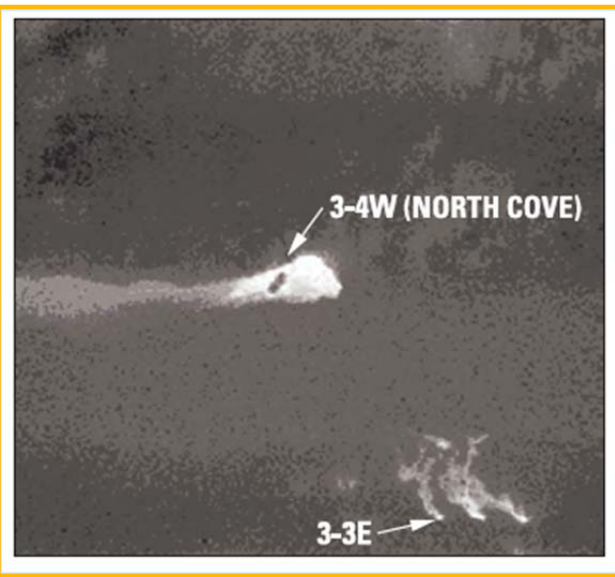




# West Branch Canal Creek, Natural Attenuation Study Area: VOCs



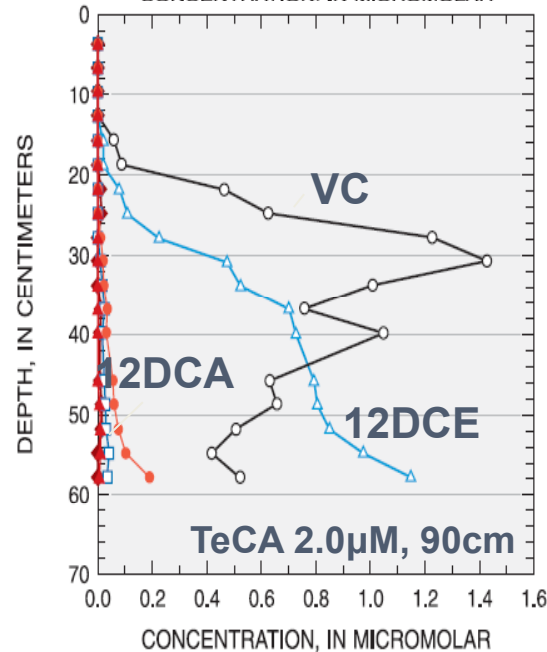
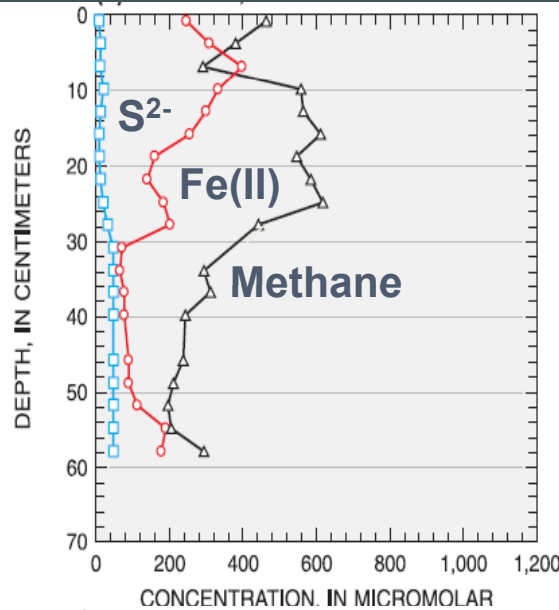
# Seep Areas



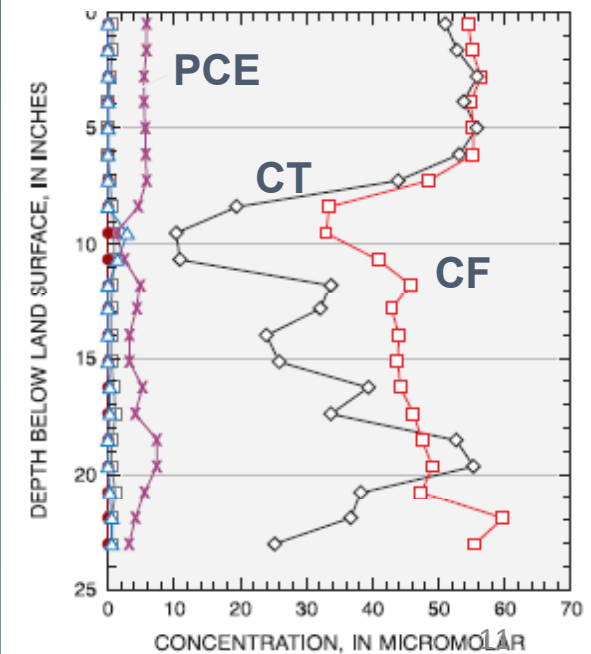
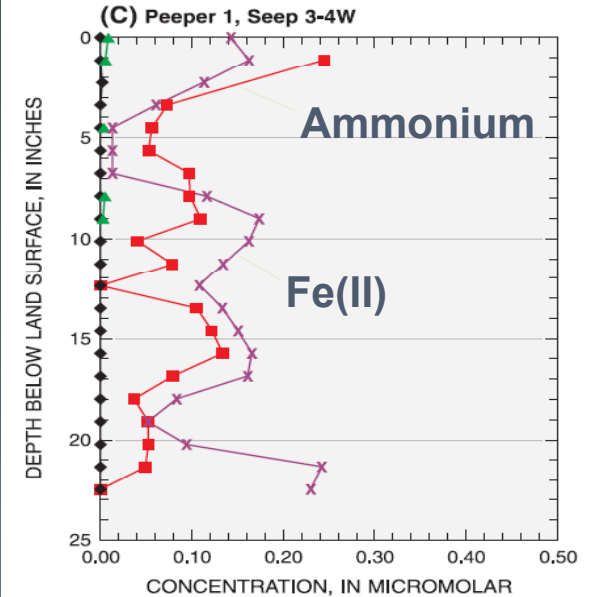
# West Branch Canal Creek

Degradation in non-seep areas where relatively slow flow allows strongly reducing conditions.

## NON-SEEP AREA

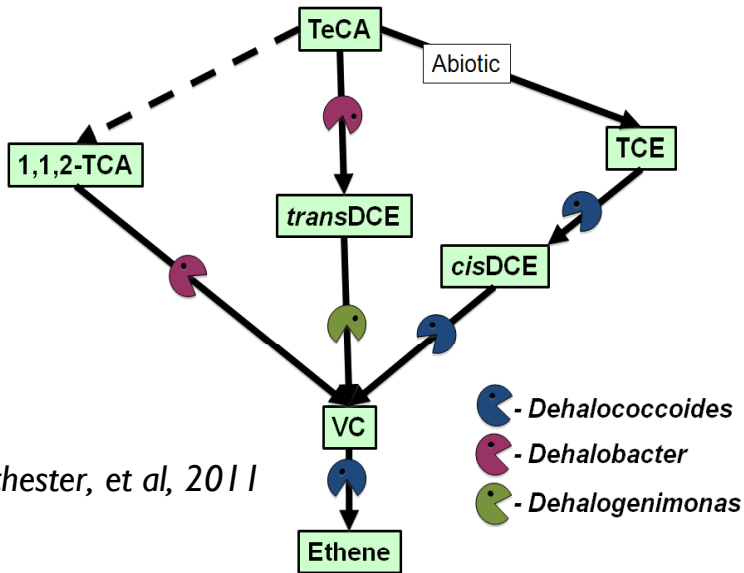
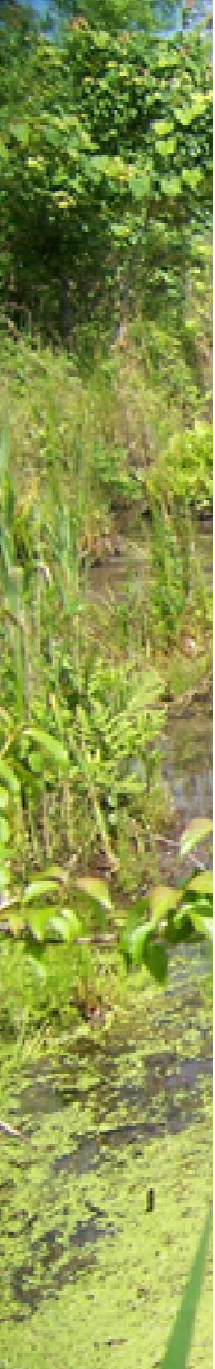


## SEEP AREA



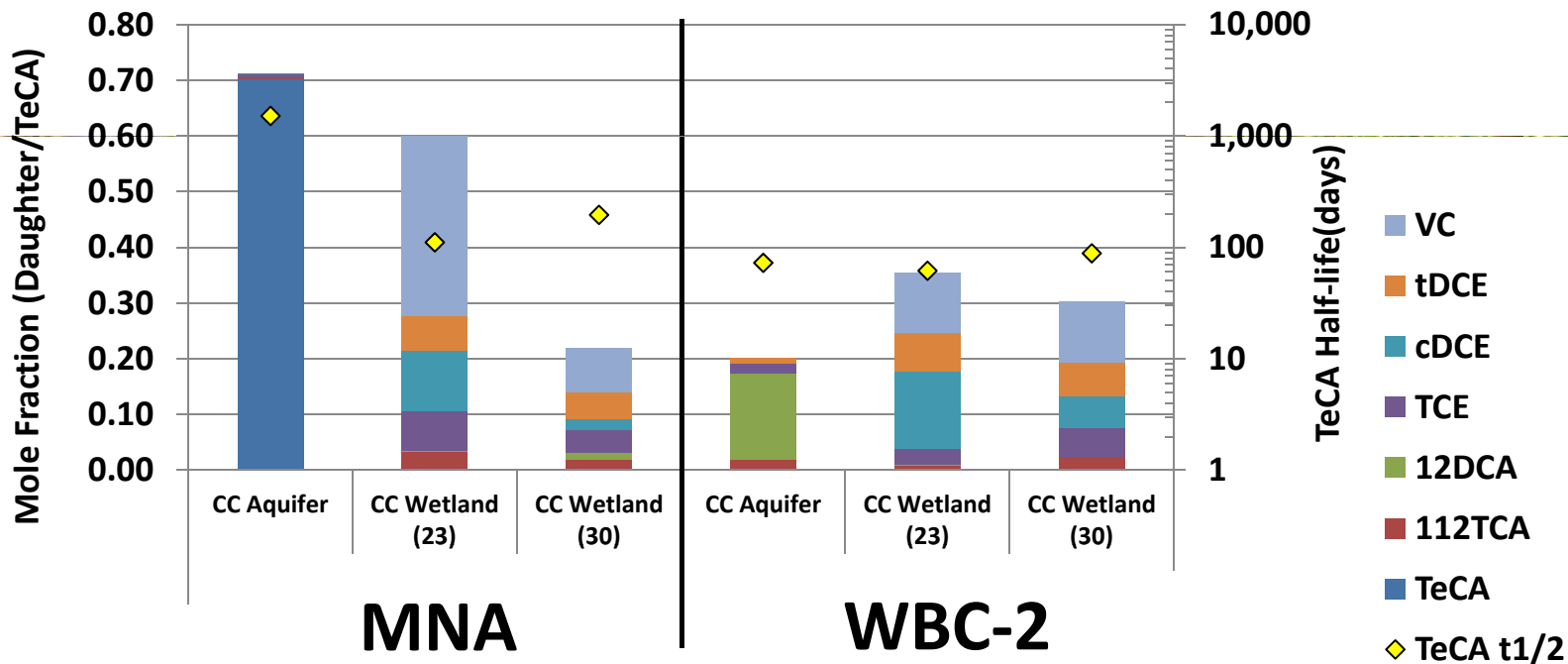
Upward flow





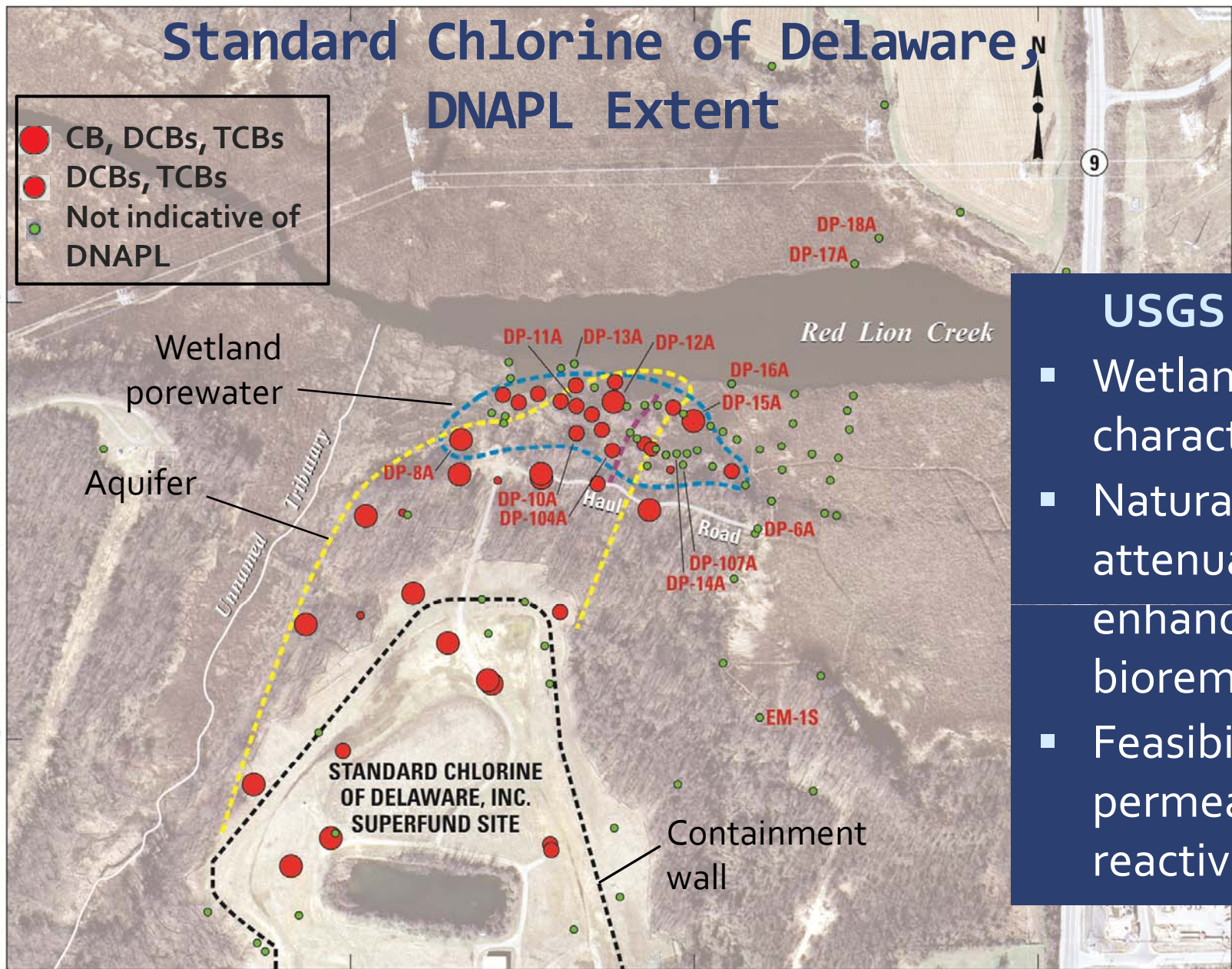
Manchester, et al, 2011

**WBC-2 Dechlorinating Consortium, developed to degrade 1,1,2,2-tetrachloroethane (TeCA)**



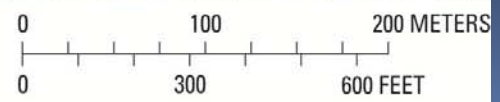
# Standard Chlorine of Delaware, DNAPL Extent

- CB, DCBs, TCBs
- DCBs, TCBs
- Not indicative of DNAPL



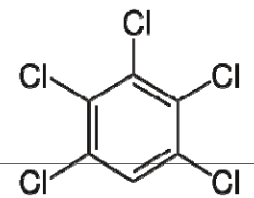
- ## USGS Study
- Wetland characterization
  - Natural attenuation; enhanced bioremediation
  - Feasibility of permeable reactive barrier

Imagery from U.S. Geological Survey, Delaware Valley Regional Planning Commission, and Department of Homeland Security, 2010, NAD 1983 State Plane Delaware Transverse Mercator Projection



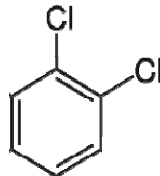
# Chlorobenzenes- Standard Chlorine of Delaware

- Anaerobic (reductive dechlorination)
  - CB serves as terminal electron acceptor
  - Separate e- donor required
  - rate decreases with decreasing number Cl
- Aerobic (oxidation)
  - O<sub>2</sub> required as electron acceptor
  - CBs utilized as C and e donor
  - Short-lived intermediates
  - rate increases with decreasing number Cl



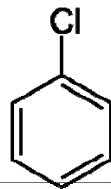
**70 (124TCB)**

**Trichlorobenzenes\***  
135TCB, 124TCB, 123TCB



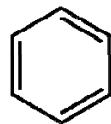
**75 (14DCB)**

**Dichlorobenzenes\***  
14DCB, 13DCB, 12DCB



**100**

**Chlorobenzene\***



**5**

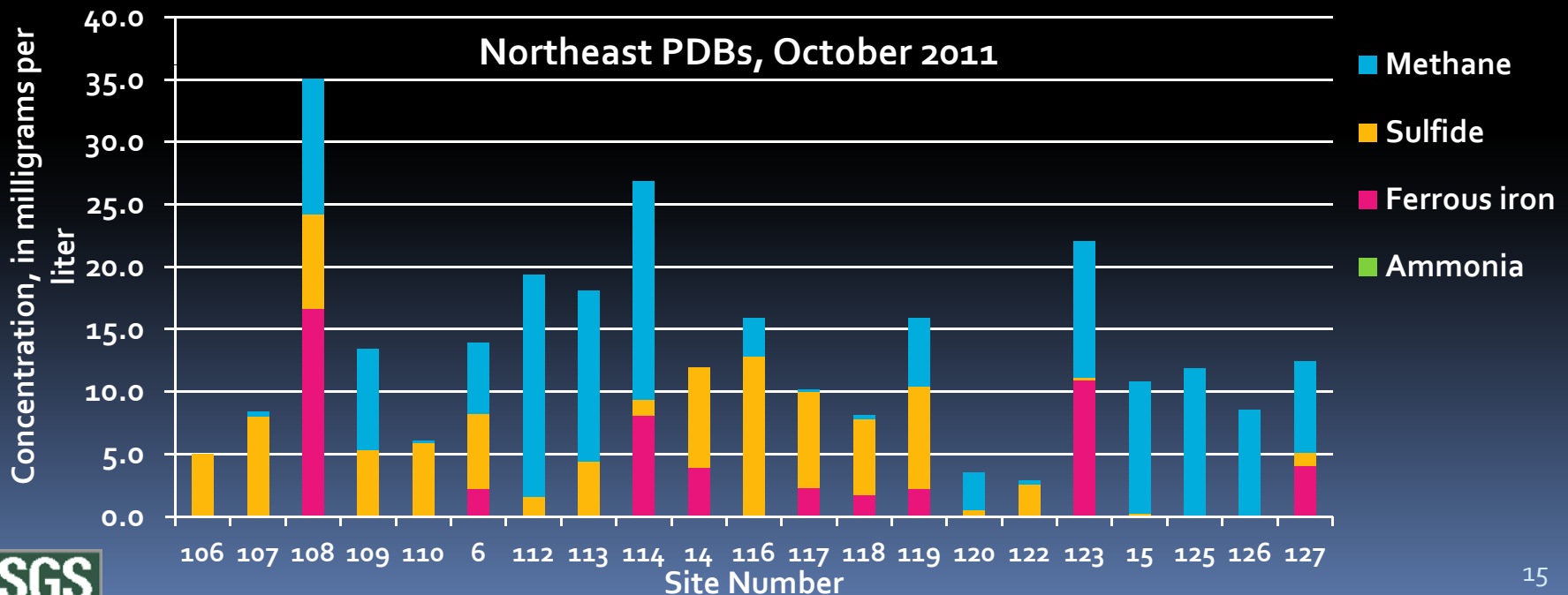
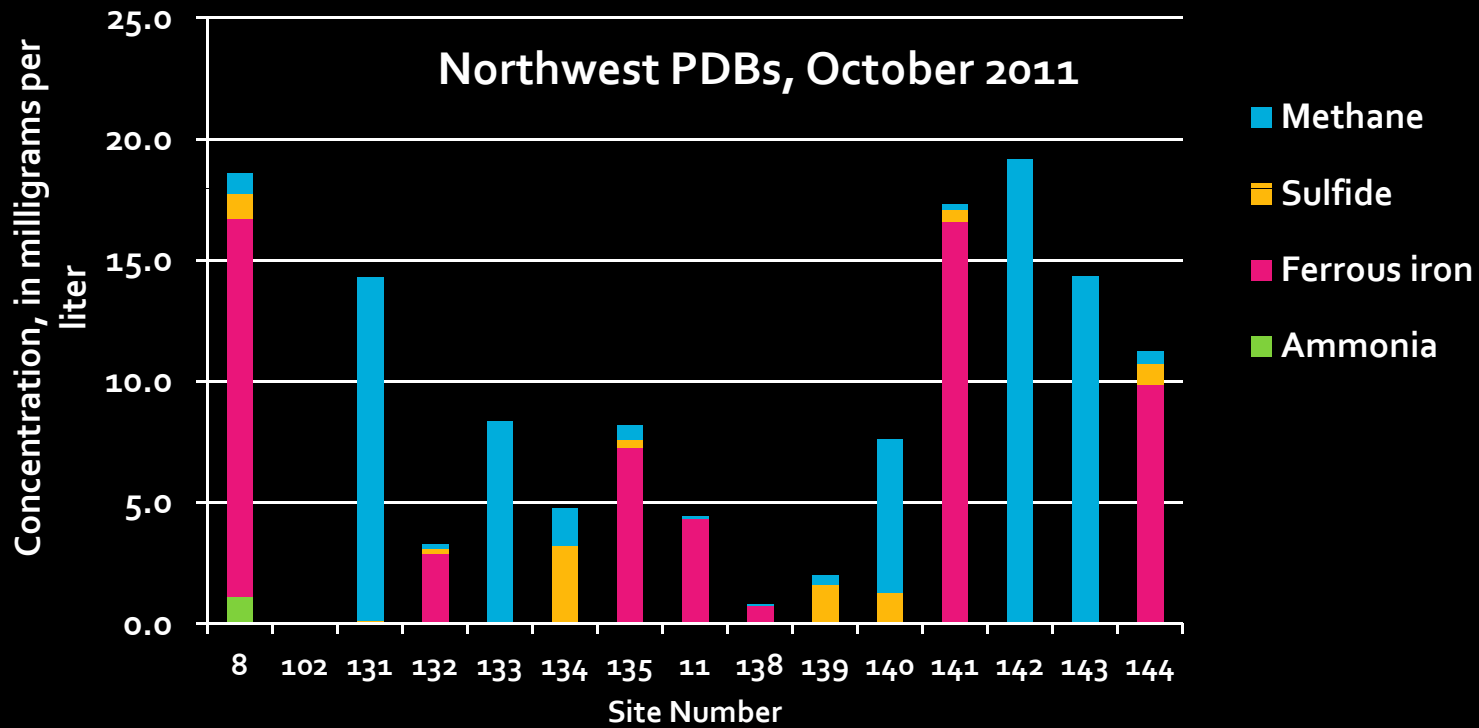
**Benzene\***

**Drinking Water  
MCL µg/L**

**CO<sub>2</sub>, CH<sub>4</sub>**

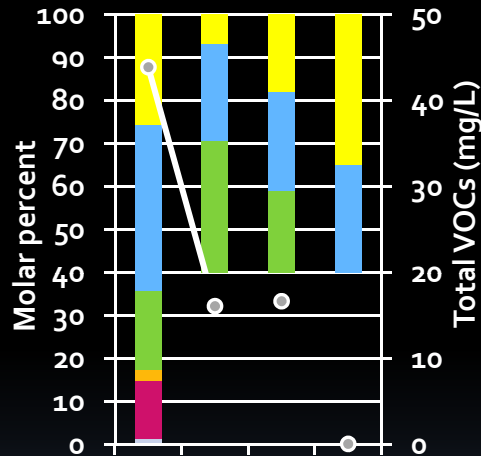
**Aerobic  
CO<sub>2</sub>,  
HCl**

\* Parent contaminant

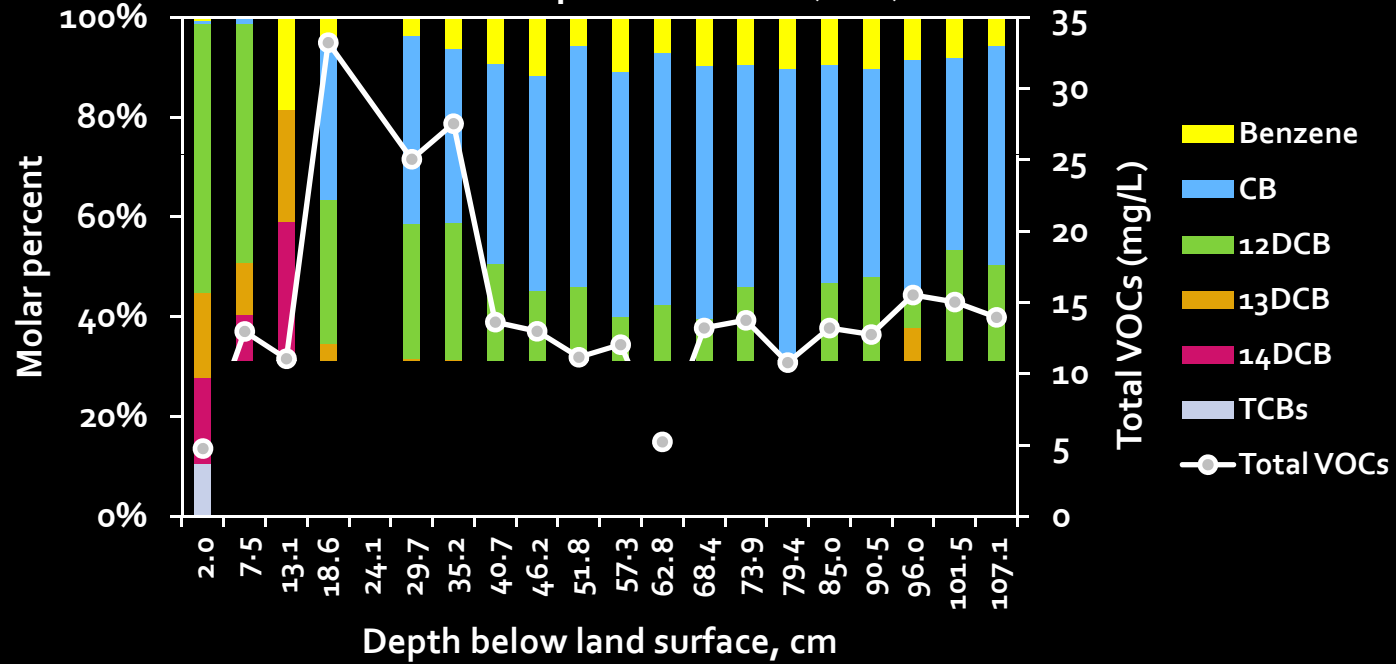


# SCD, VOCs in Peepers

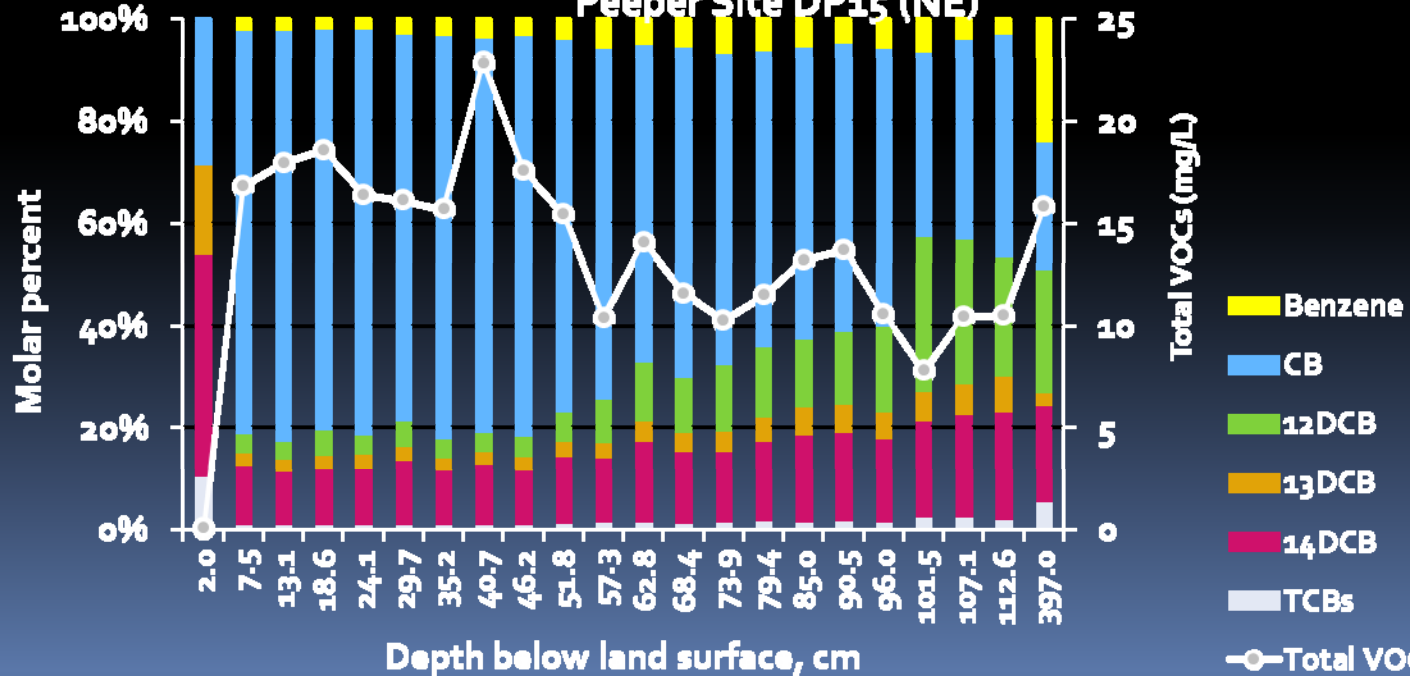
Upland Wells, Oct. 2011



Peeper Site DP-8 (NW)



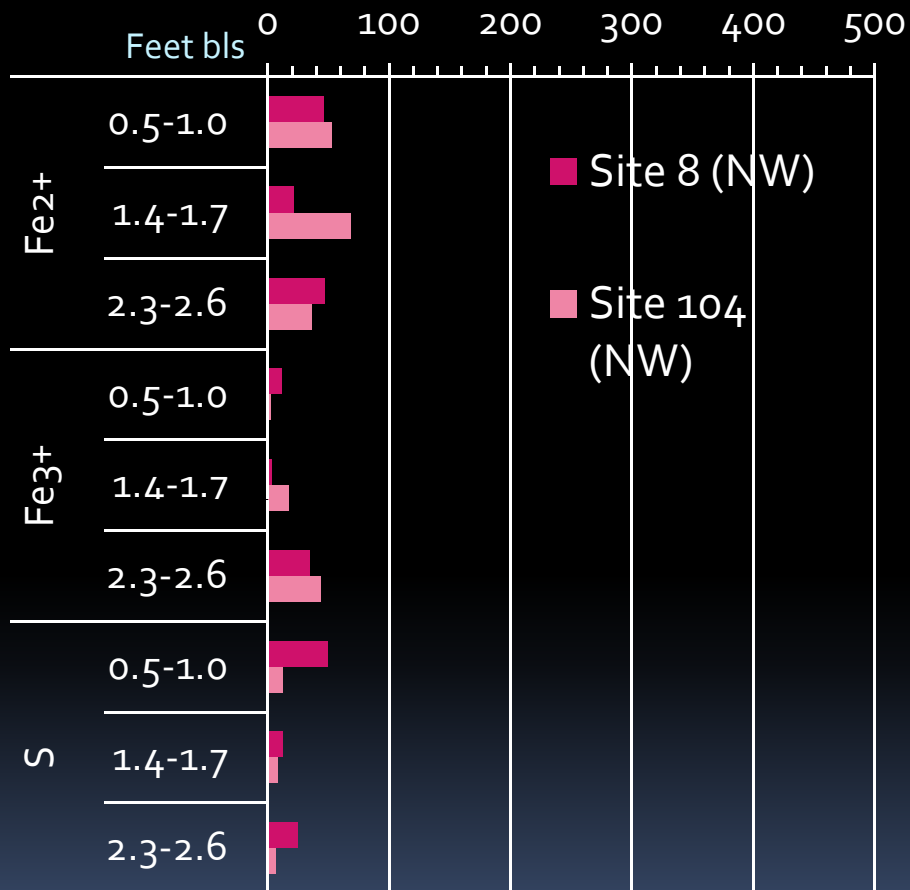
Peeper Site DP15 (NE)



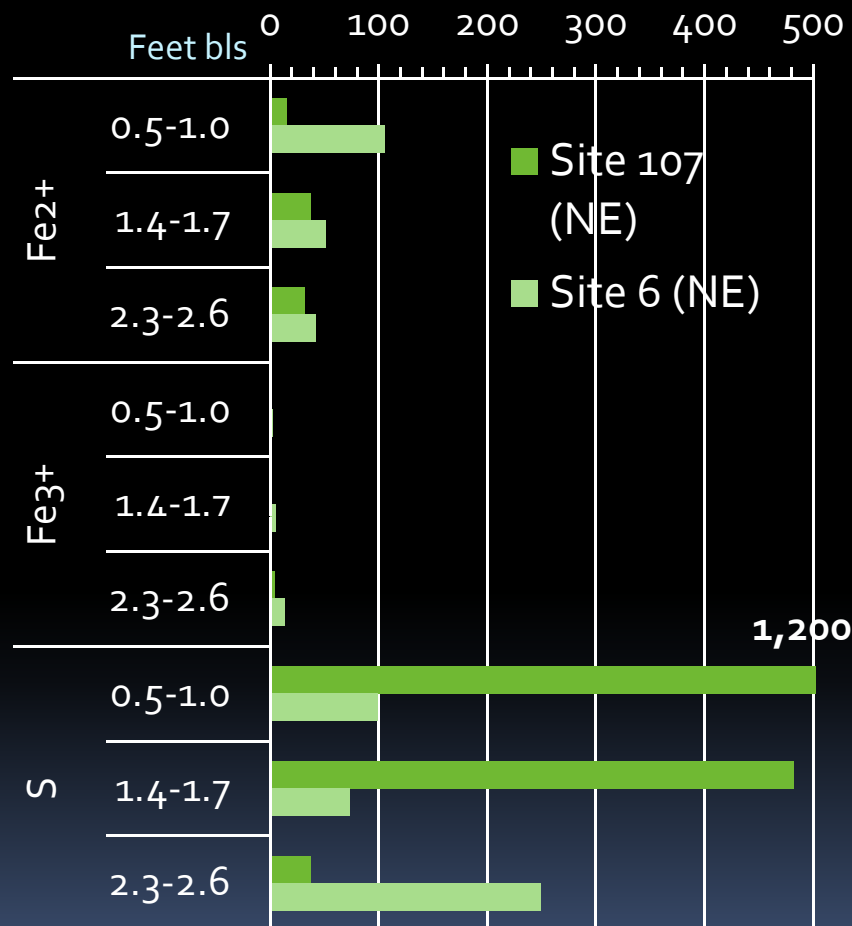


# Iron, Sulfide- Sediment Cores

Concentration,  $\mu\text{mol/g}$

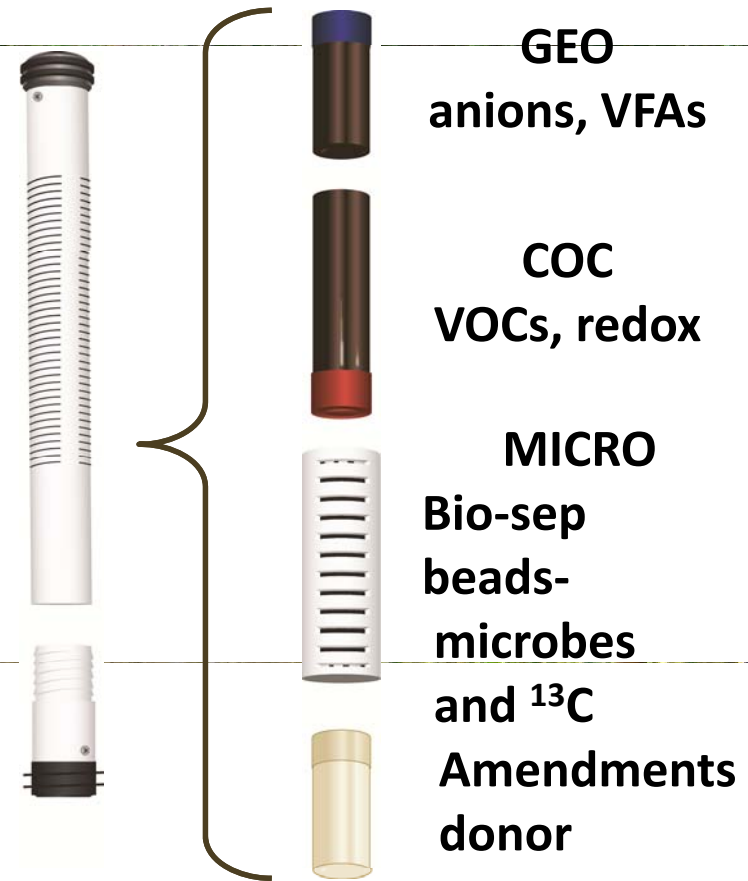


Concentration,  $\mu\text{mol/g}$

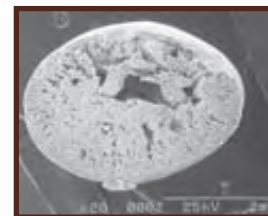


# Tools to Evaluate Biodegradation

- **Molecular Biological Tools**
  - Quantitative PCR- Counts genes, taxonomic or functional, for specific targets; micro-arrays (QuantArray, MI)
  - Terminal Restriction Fragment Length Polymorphisms (TRFLP)- fingerprint of the microbial community
  - Next-generation sequencing (high throughput) - in depth profile of the microbial community; Illumina, 454 sequencing
- **Stable Isotopes**
  - SIP, Stable Isotope Probing-  $^{13}\text{C}$  used as a tracer
  - CSIA, Compound Specific Isotope Analysis- isotopic fractionation in parent and metabolites



## Bio-Traps, Microbial Insights

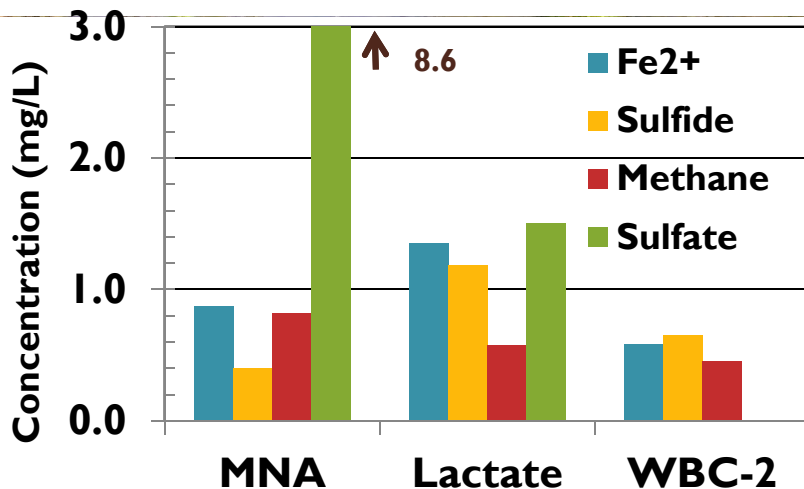


Bio-Sep® beads provide a large surface area for microbial attachment

# In situ microcosms with Bio-Traps (Microbial Insights)



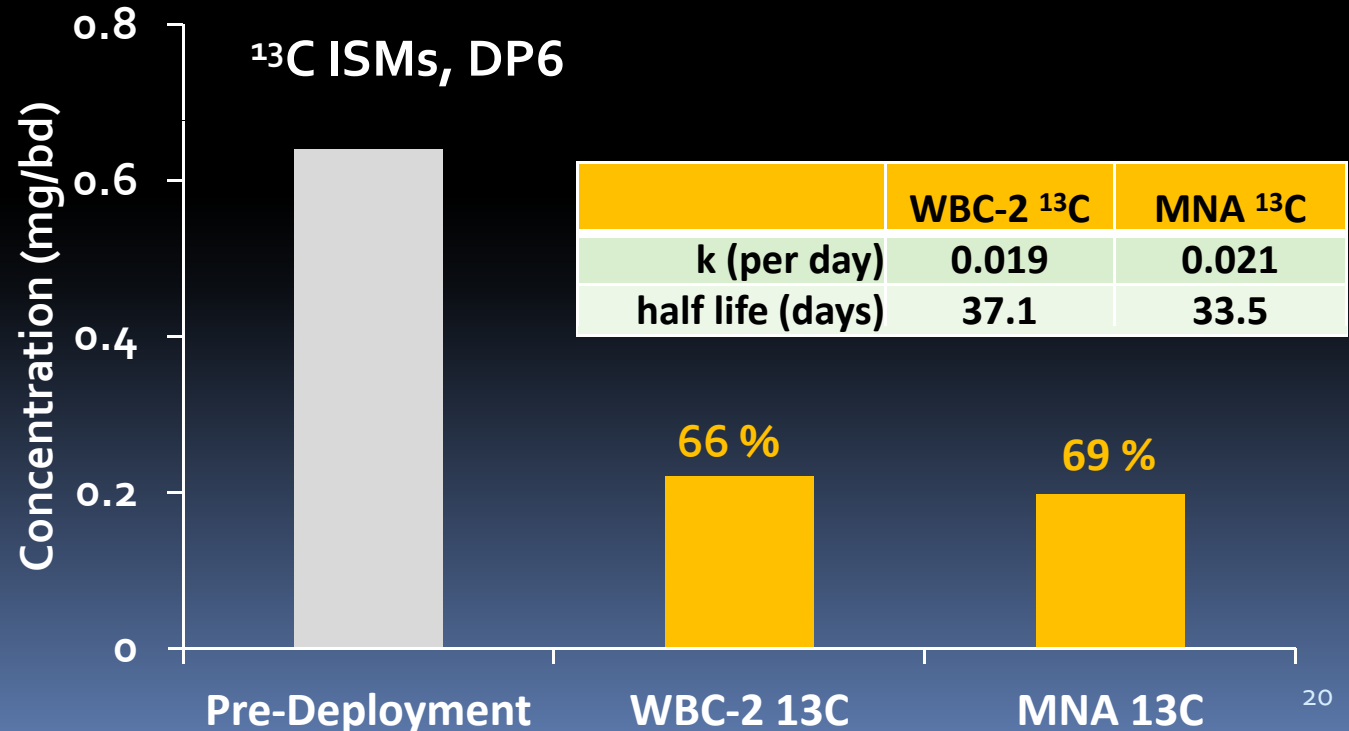
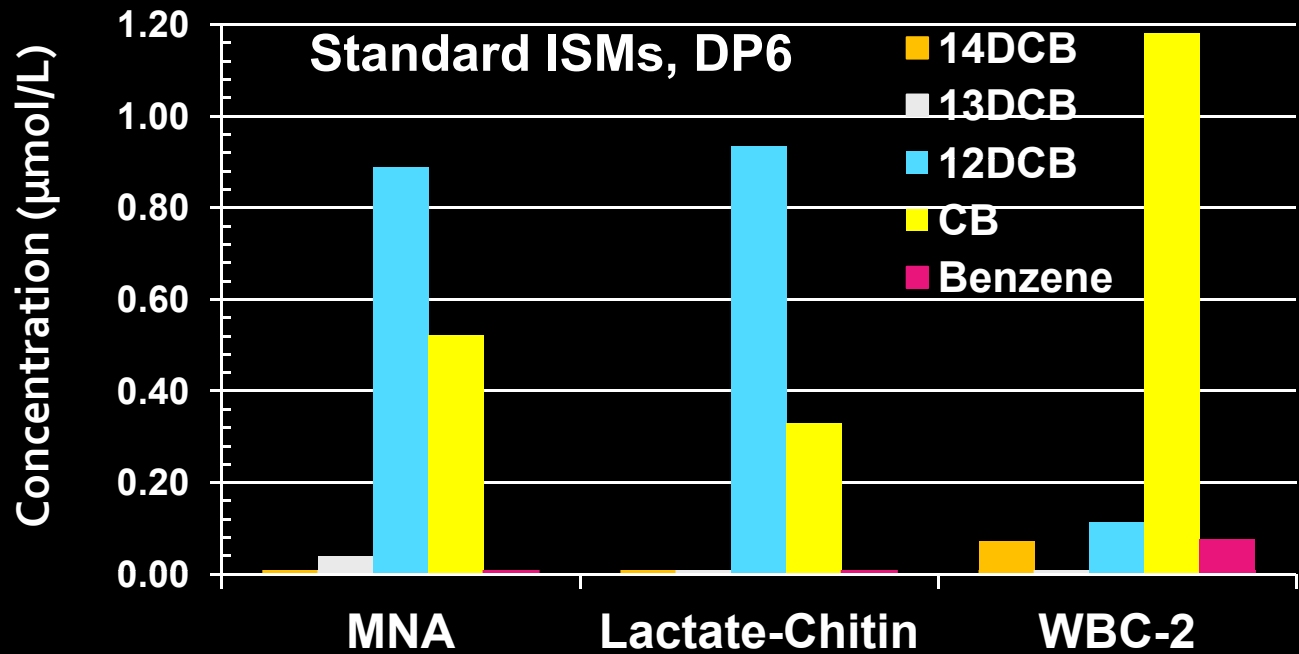
Site 107 Redox



- Two each northwest and northeast sites
- Three standard treatments and three <sup>13</sup>C-labeled treatments
  - MNA, monitored natural attenuation (no amendments)
  - Lactate, biostimulated with lactate + chitin
  - WBC-2, bioaugmented
- <sup>13</sup>C-labeled chlorobenzene
- QuantArray analysis of species and functional genes for aerobic and anaerobic biodegradation

# ISM Results:

- Complete degradation of DCBs evident in WBC-2 treatment in standard ISMs
- $^{13}\text{C}$ -labeled ISMs showed complete degradation of monochloro-benzene in MNA and WBC-2

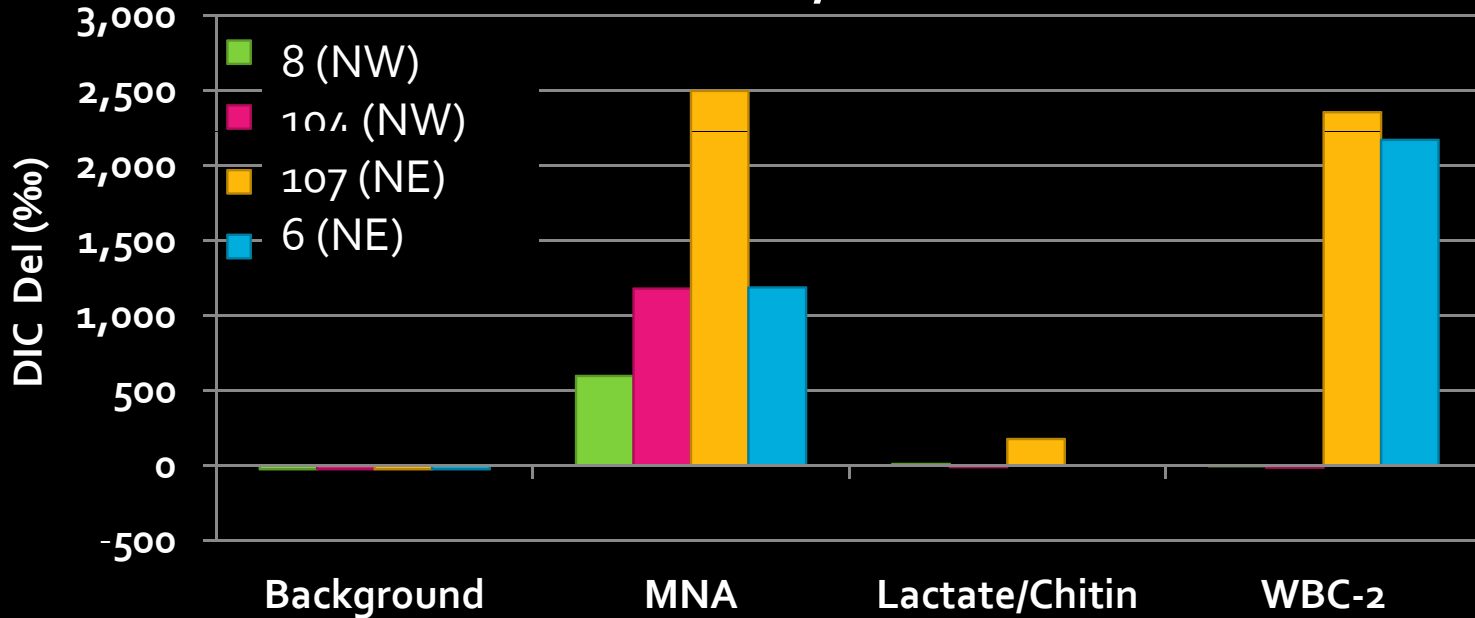


# Bio-Traps: <sup>13</sup>C-labeled Chloro- benzene

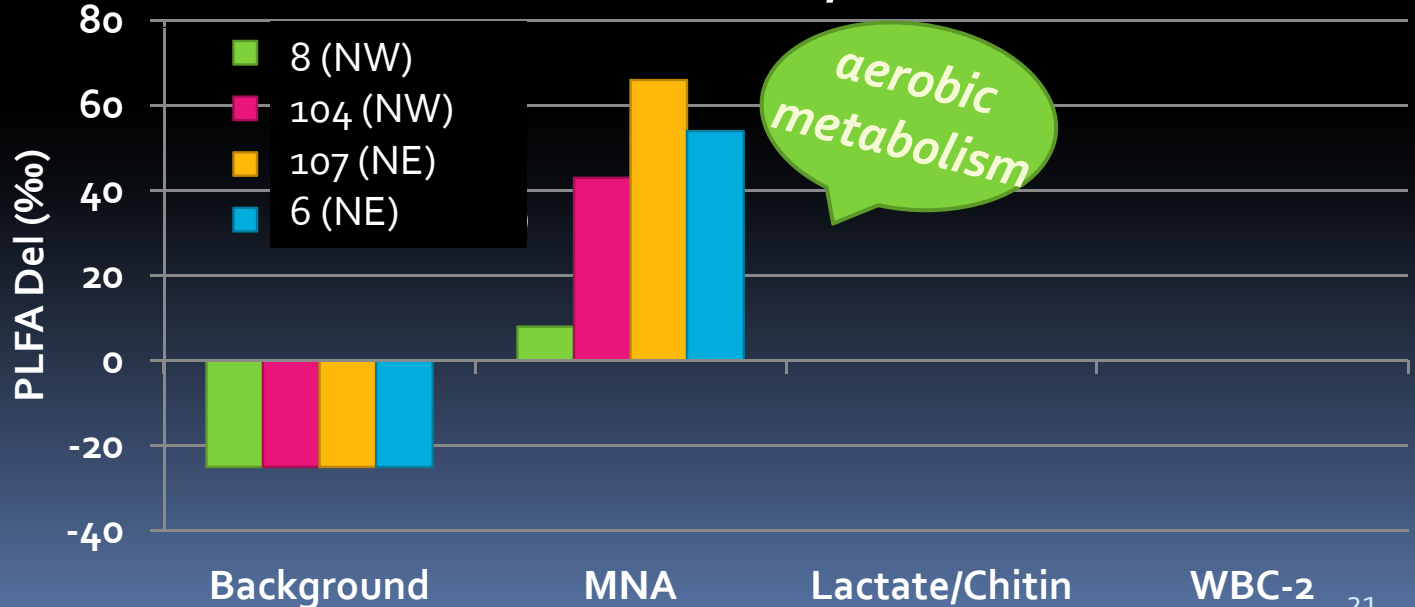
Incorporation in  
dissolved  
inorganic carbon  
= Mineralization

Incorporation  
in PLFA =  
Metabolism  
(C for growth)

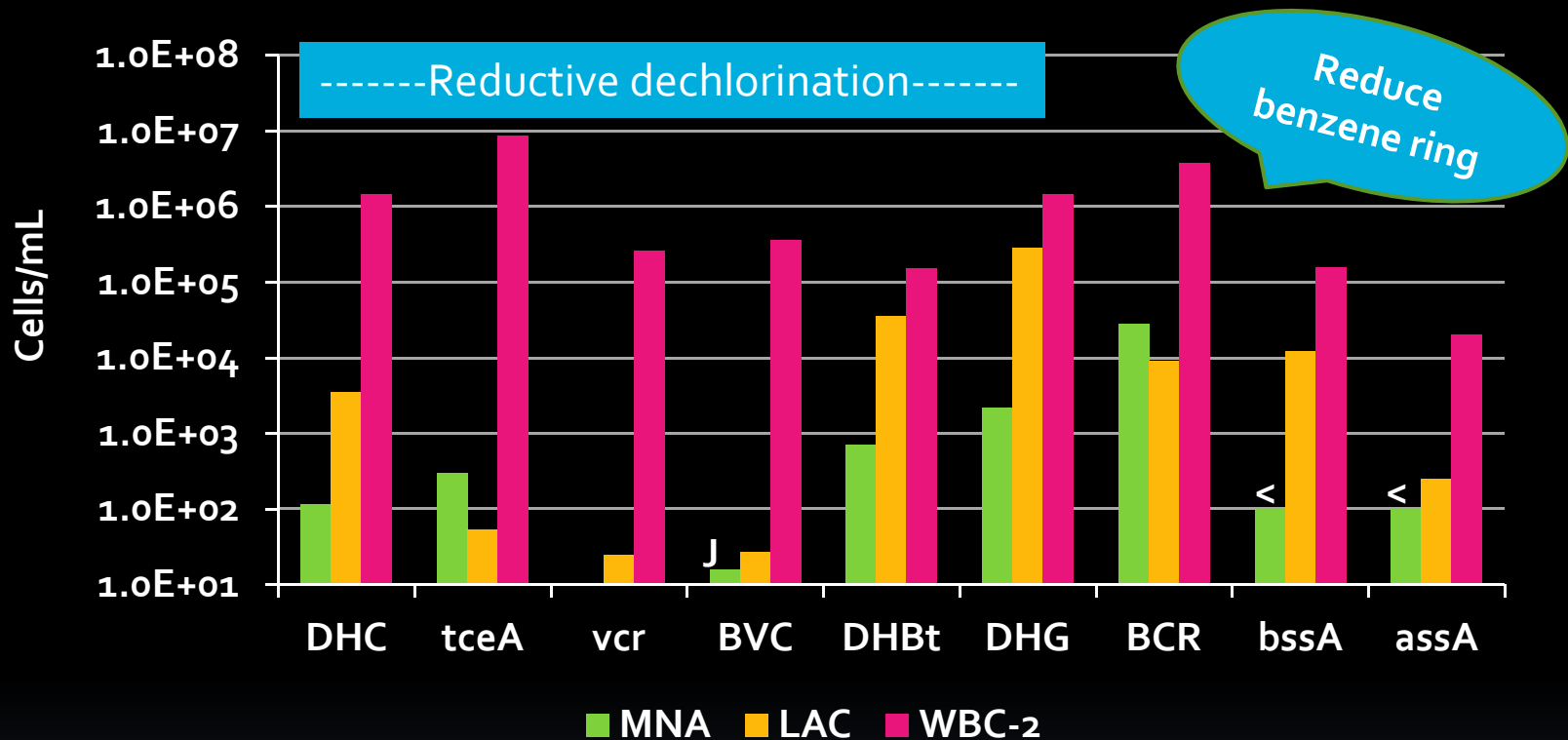
## <sup>13</sup>C Utilized for CO<sub>2</sub>, <sup>13</sup>C Chlorobenzene



## <sup>13</sup>C Utilized for Biomass, <sup>13</sup>C Chlorobenzene



# QuantArray Microbial Analysis- Anaerobic



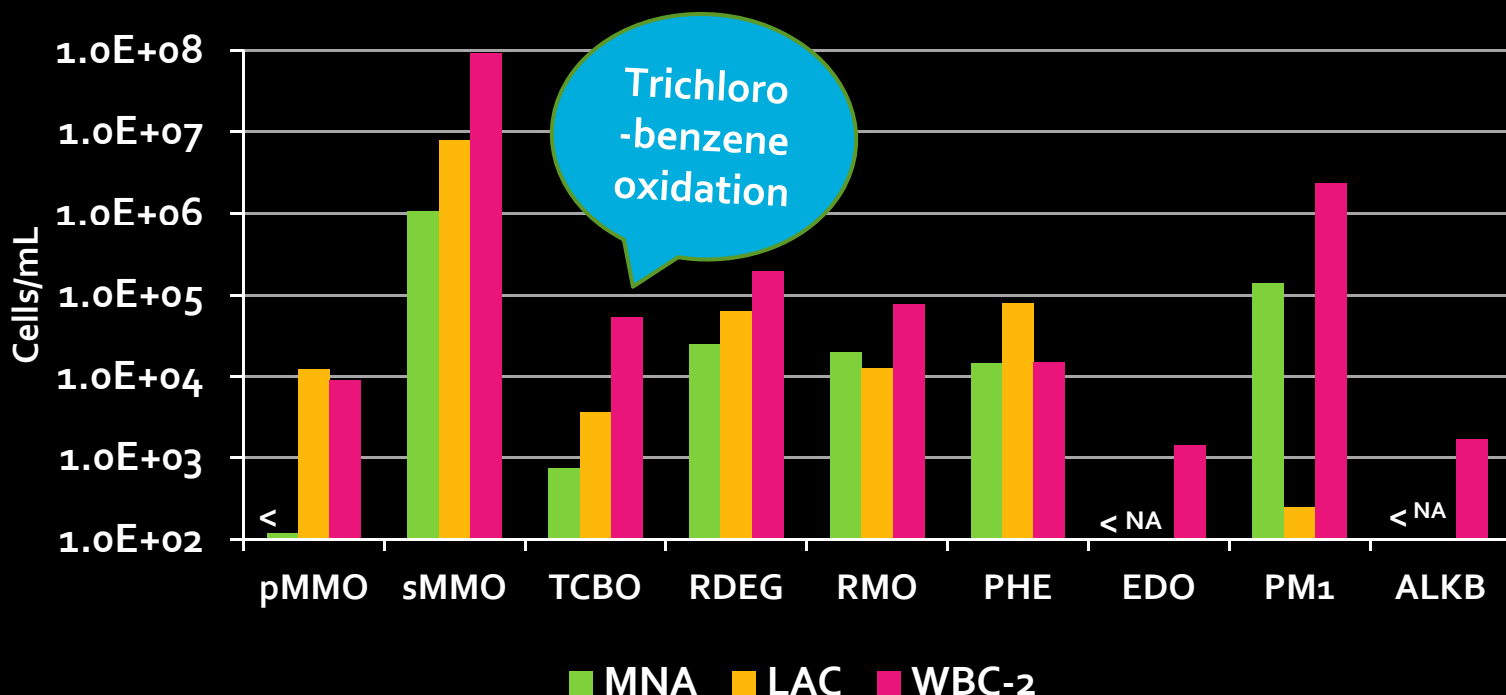
## Reductive dechlorination:

DHC , Dehalococcoides spp.  
 TCE, tceA reductase  
 VCR, vinyl chloride reductase  
 BV<sub>1</sub> , vinyl chloride reductase  
 DHBt, Dehalobacter spp.  
 DHG, Dehalogenimonas spp.

## BTEX, PAHs and alkanes:

BCR, Benzoyl coenzyme A reductase  
 bssA, benzylsuccinate synthase  
 assA, alkylsuccinate synthase

# QuantArray Microbial Analysis- Aerobic



pMMO, particulate methane monooxygenase  
 sMMO, soluble methane monooxygenase  
 TCBO, trichlorobenzene dioxygenase  
 RDEG, toluene monooxygenase 2  
 RMO, toluene monooxygenase

PHE, phenol hydroxylase  
 EDO, ethylbenzene/isopropylbenzene dioxygenase  
 PM<sub>1</sub>, Methylibium petroliphilum PM<sub>1</sub>  
 ALKB, alkane monooxygenase

# Changing Paradigm

Previous paradigm for chlorinated VOCs:

- Anaerobic reductive dechlorination only process in apparent low redox zones
- Aerobic oxidation requires measurable oxygen
- Anaerobic oxidation responsible for losses of lower VOCs at anaerobic plume fringes

*Perils of Categorical Thinking: "Oxic/Anoxic" Conceptual Model in Environmental Remediation*

Bradley 2012

*Microbial Mineralization of Dichloroethene and Vinyl Chloride under Hypoxic Conditions*

Bradley and Chapelle 2011

*Isolation of an aerobic vinyl chloride oxidizer from anaerobic groundwater*

Fullerton et al. 2014

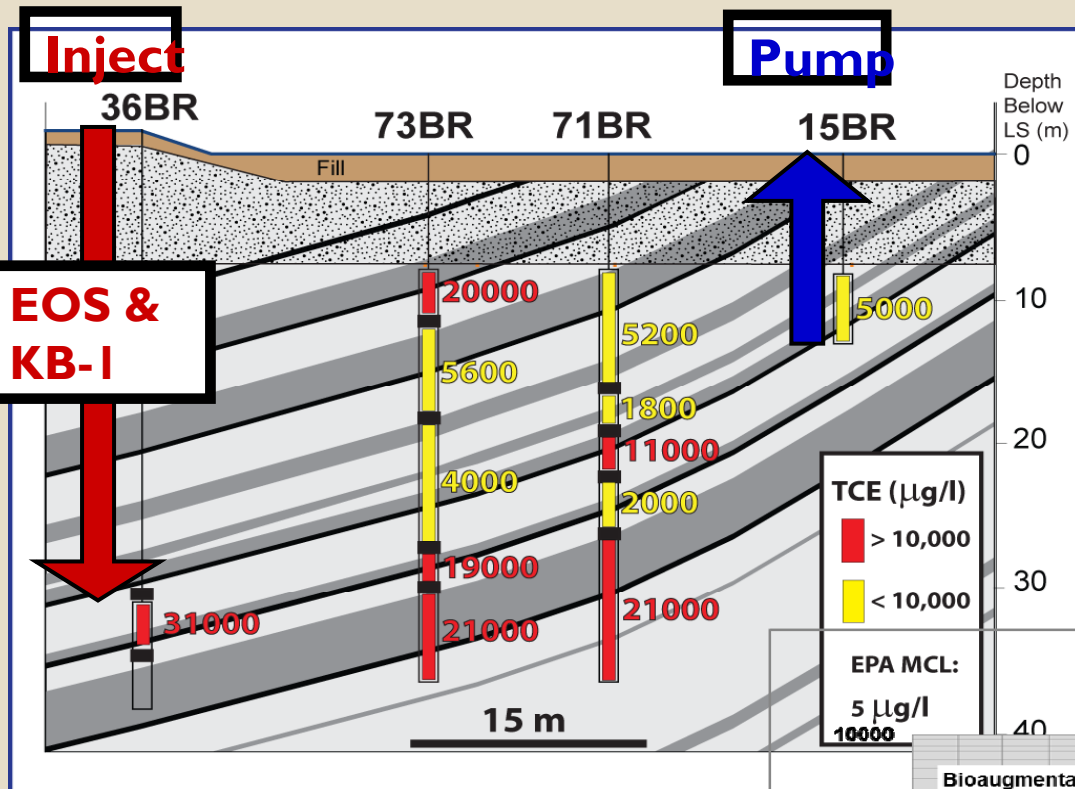
*Sustained Aerobic Oxidation of Vinyl Chloride at Low Oxygen Concentrations*

Gossett 2010

*Concurrent and Complete Anaerobic Reduction and Microaerophilic Degradation of Mono-, Di-, and Trichlorobenzenes*

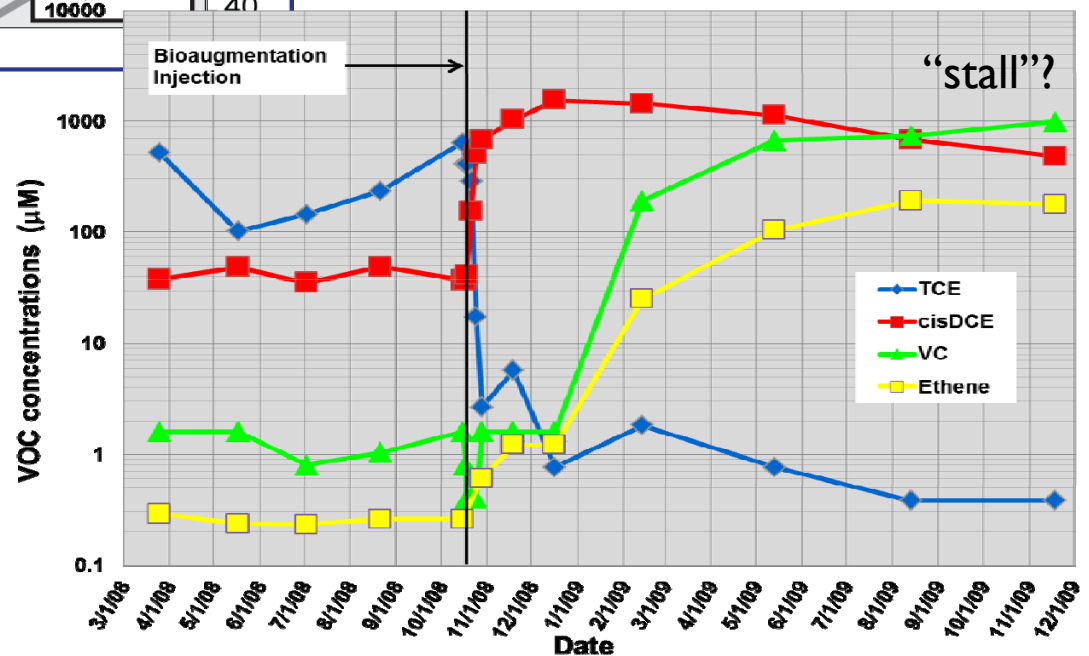
Burns et al. 2013





Bioaugmentation:  
Fractured  
sedimentary rock  
aquifer, former  
Naval Air Warfare  
Center (NAWC)

**VOCs vs Time**  
Injection Well - 36BR-A

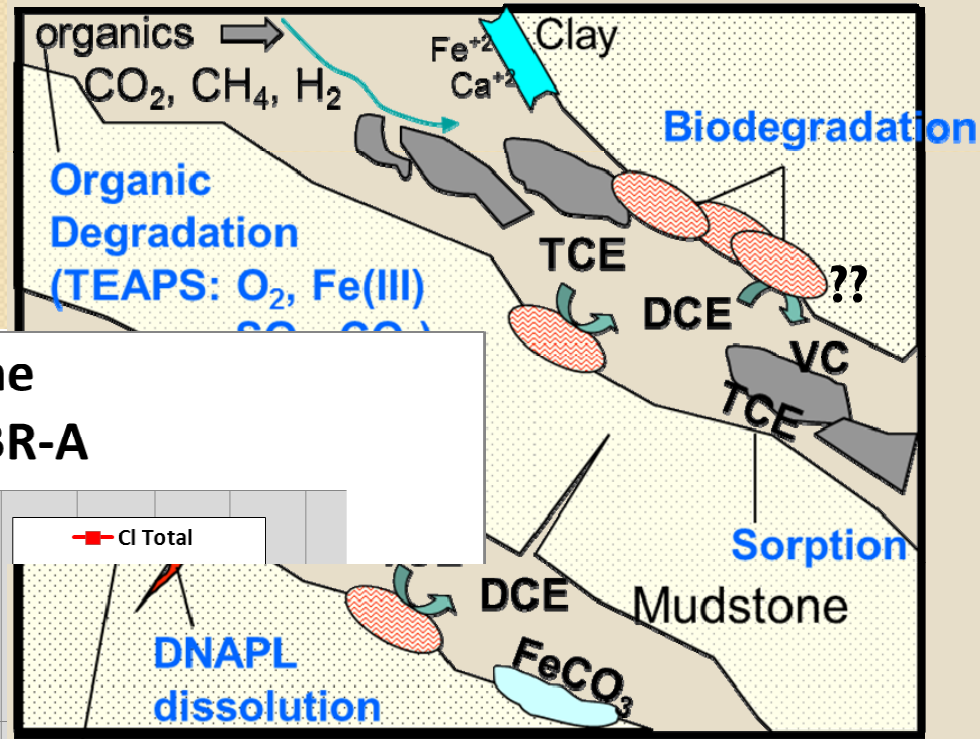


Toxic Substances Hydrology  
Program

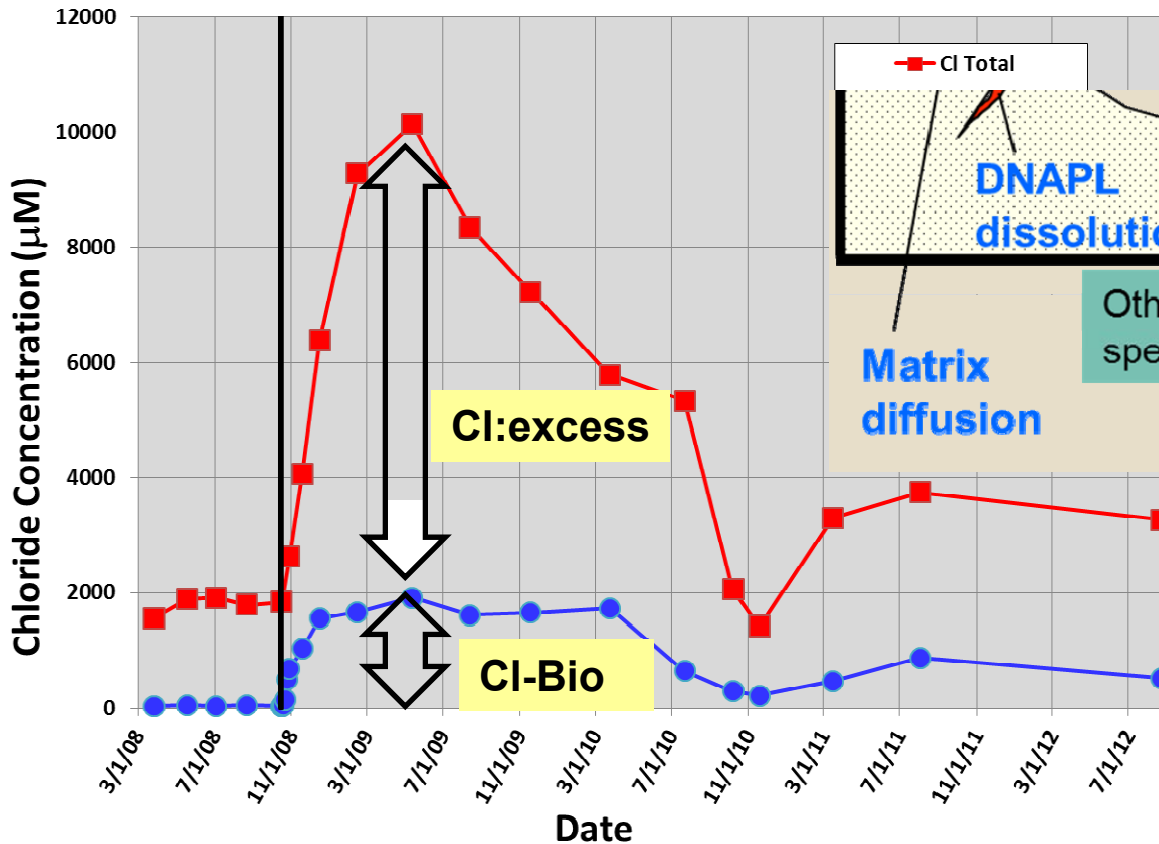
New Jersey Water Science Center  
National Research Program



# Matrix diffusion/ DNAPL dissolution



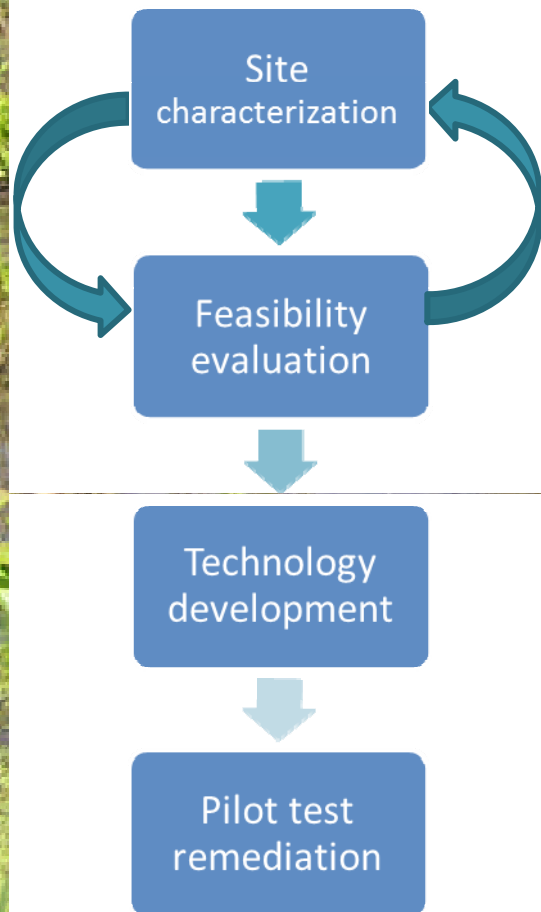
**Chloride vs Time  
Injection Well 36BR-A**



(Modified from Tom Imbrigiotta)

Currently investigating changes in native and bioaugmented microbial communities- toxicity/inhibition effects cause growth of “partial dechlorinators”?

# Use of laboratory testing to characterize microbial communities and biodegradation processes



Native Bioreactor



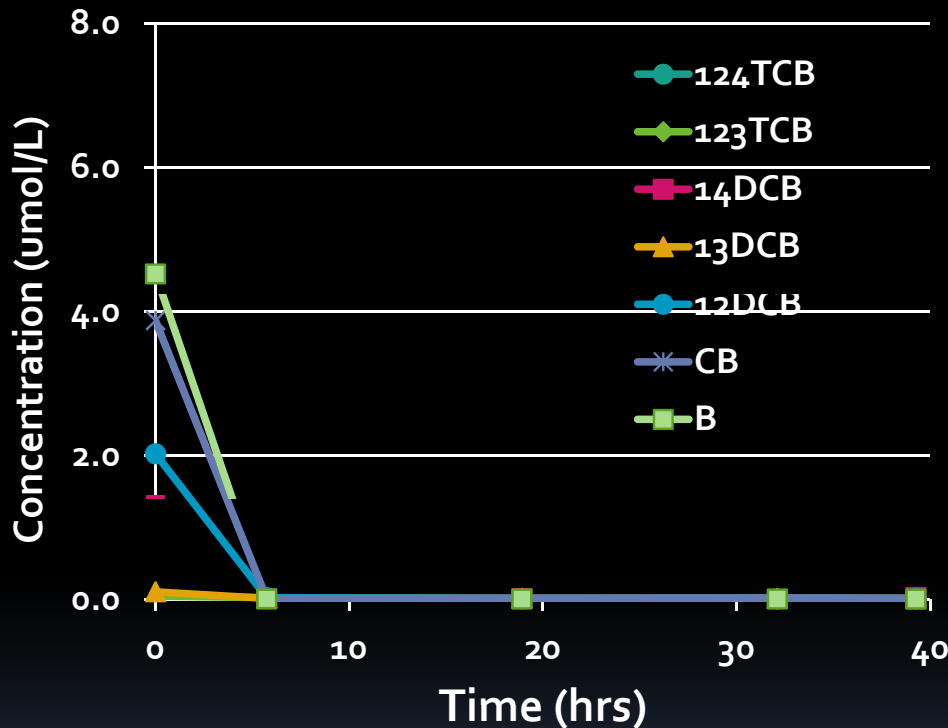
Bioaugmented Bioreactor



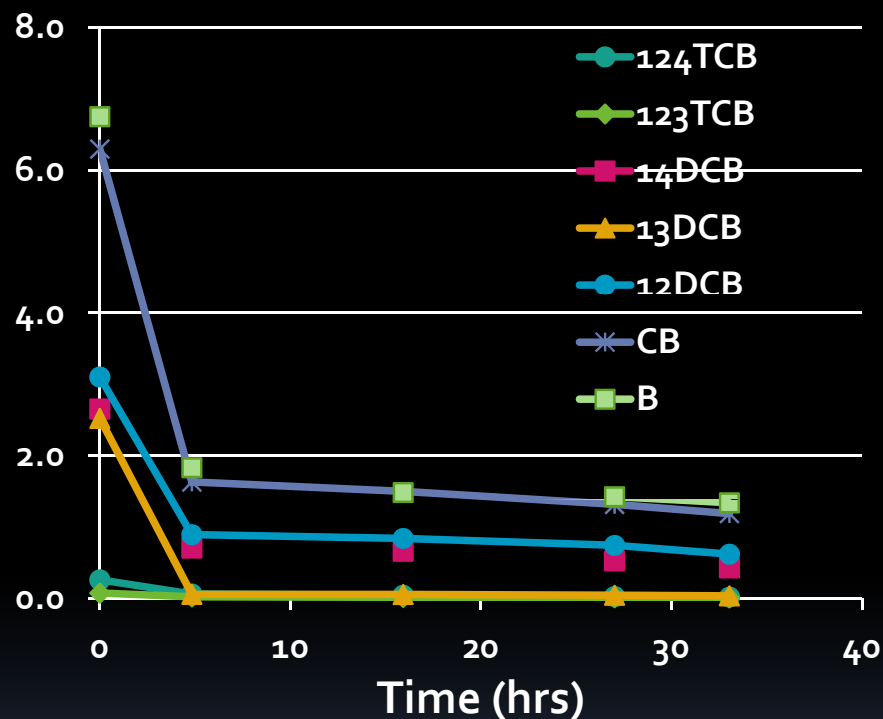
Polyethylene and polyurethane support matrix for building biofilm of native or bioaugmented microorganisms

# SCD Bioreactors- Aerobic vs. Anaerobic

C Reactor 8/13/12  
aerobic



D Reactor 8/10/12  
anaerobic



- *occurrence of aerobic and anaerobic degradation by native bacteria*
- *aerobic degradation faster than anaerobic biodegradation*
- *WBC-2 able to degrade chlorinated benzenes and benzene anaerobically*
- *accumulation of daughter products not evident*

Questions?

7.0 kV X10.0k 3.00µm

# Acknowledgements

## USGS MD-DE-DC

### Fate and Bioremediation Team

Michelle Lorah

Jessica Teunis

Mastin Mount

Michael Brayton

Charles Walker

Roberto Cruz

Emily Majcher

Anna Baker

Luke Myers

## NRP Reston Collaborators

Isabelle Cozzarelli

Denise Akob

Allen Shapiro

Dan Goode

Tom Imbrigiotta

Claire Tiedeman

## CRADA-

### Geosyntec Consultants

Duane Graves

## USEPA Region III



US Army, Aberdeen Proving Ground Installation Restoration Program

## Toxic Substances Hydrology Program

